

# Methods Used to Collect and Interpret Raw Data from a GAC Adsorption Pilot Plant Study to Identify Fundamental Design and Operating Criteria

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# Looking Forward

- The Setup
- Mountains of Data
- Typical Analysis
- Expanded Analysis
- Broad and Useful Interpretations

# The Setup



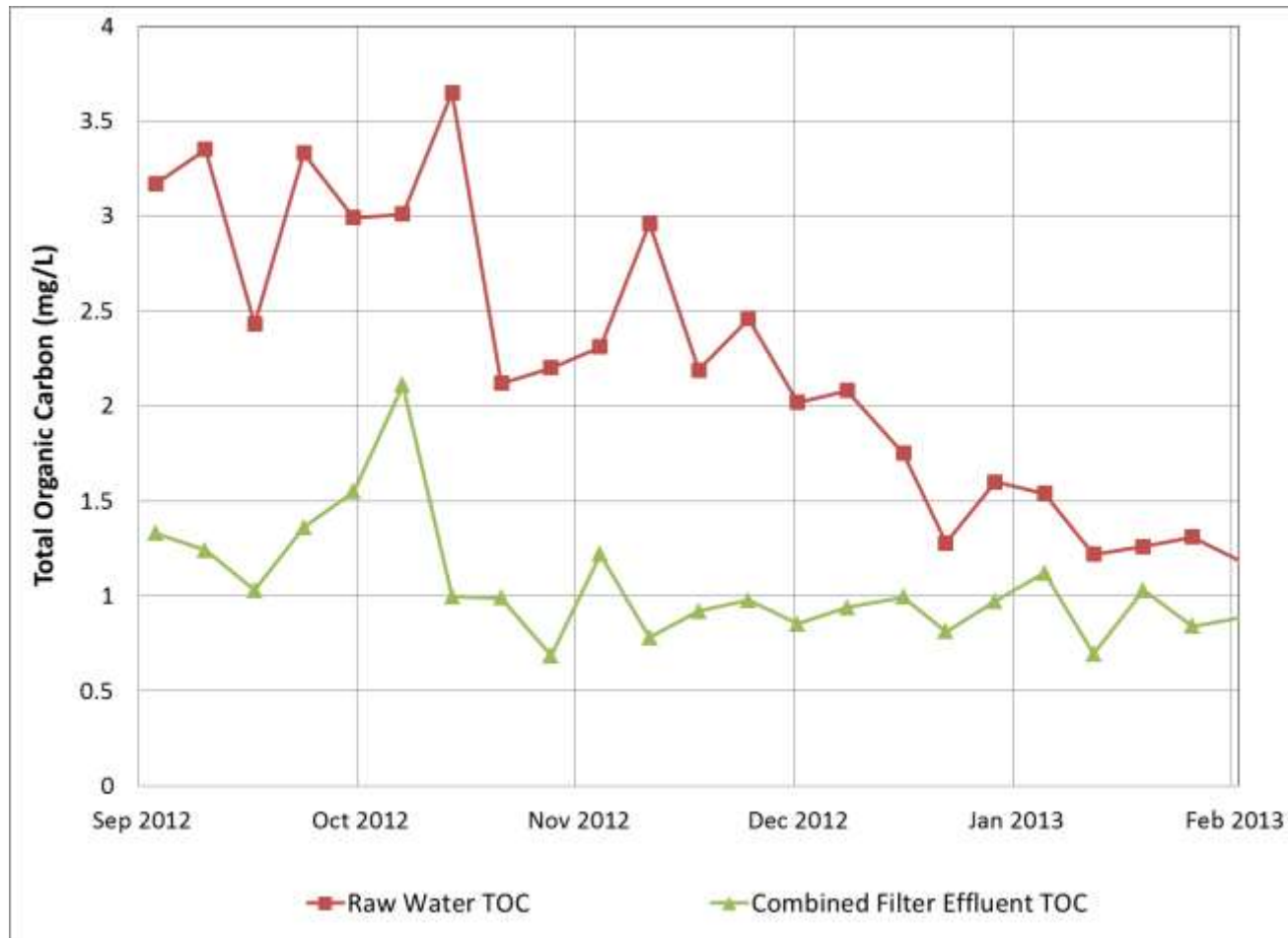
# Testing Performed on a 13 MGD Pacific Island WTP



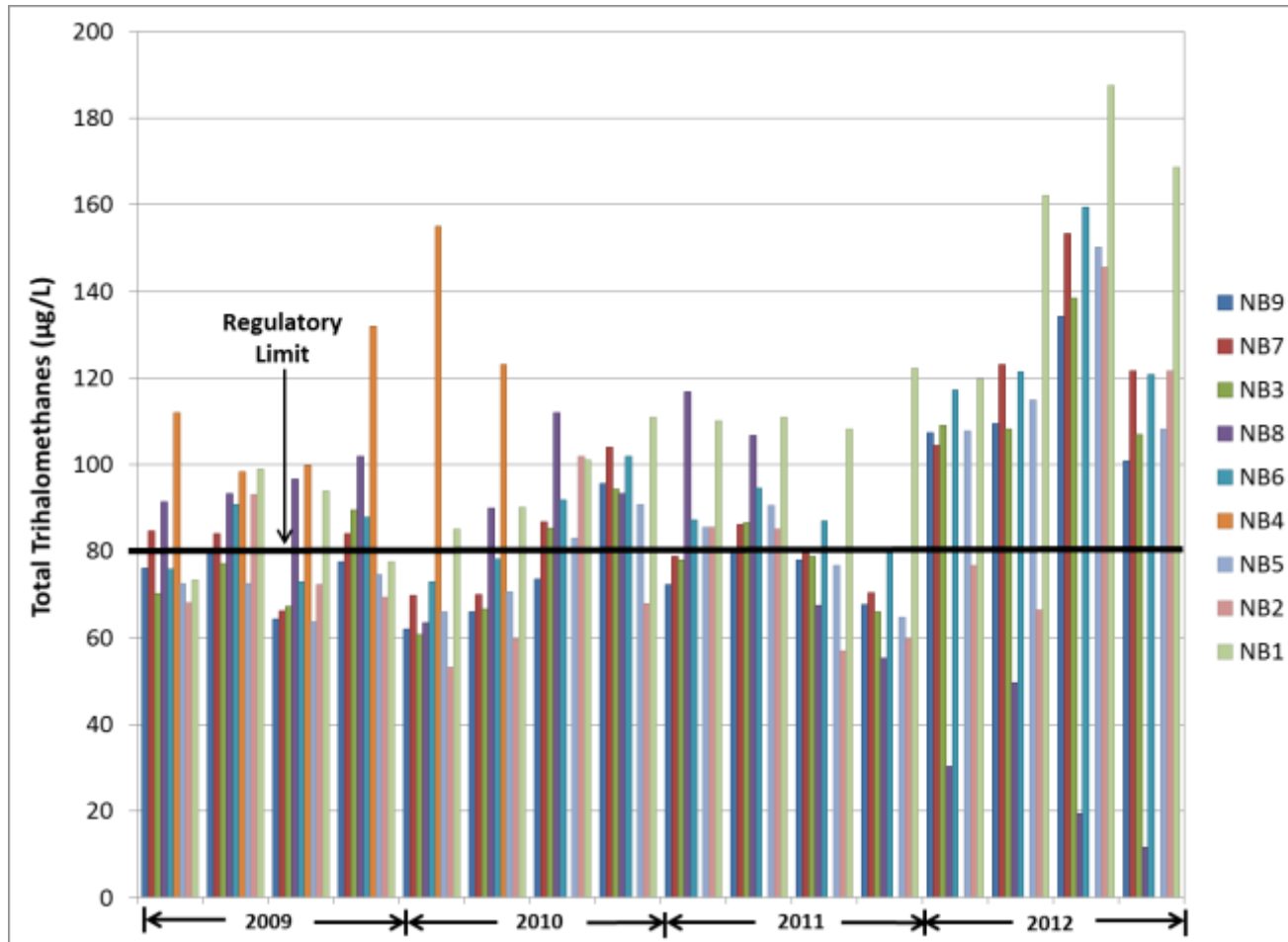
Warm Tropical Water  
Lots of Organics!



# TOC Data Seems Modest



# However, Problems with TTHM!



# GAC Adsorption is A Good Option

## DBP Reduction Alternatives

PRIMARY ALTERNATIVE	Treatment method or operational change
1	GAC adsorption
2	Magnetic ion exchange (MIEX) filtration
3	Membrane filtration
4	Chloramine disinfection
5	Powdered activated carbon (PAC) addition
6	Use of alternative oxidants for pretreatment (ozone)
SECONDARY ALTERNATIVE	Treatment method or operational change
7	Reduced chlorine dosing prior to filtration (prechlorination)
8	Reduced chlorine dosing in treated water (post-chlorination)
9	Use of alternative oxidants for pretreatment (chlorine dioxide)
10	Use of alternative oxidants for pretreatment (potassium permanganate)
11	Enhanced coagulation
12	Increased distribution system water line flushing
13	Tank aeration of DBPs (chloroform only)

# Test Program

- Isotherm Testing of Various GAC Media
- Determine Adsorber Operating Criteria
  - DOC surrogate for TTHM
  - Breakpoint
  - Breakthrough-assigned to be at half of the TTHM MCL
  - Exhaustion-equivalent to feed stream concentration
- Operate a Single Adsorber for 5 Months
- Collect and Analyze Data
- Interpret Results



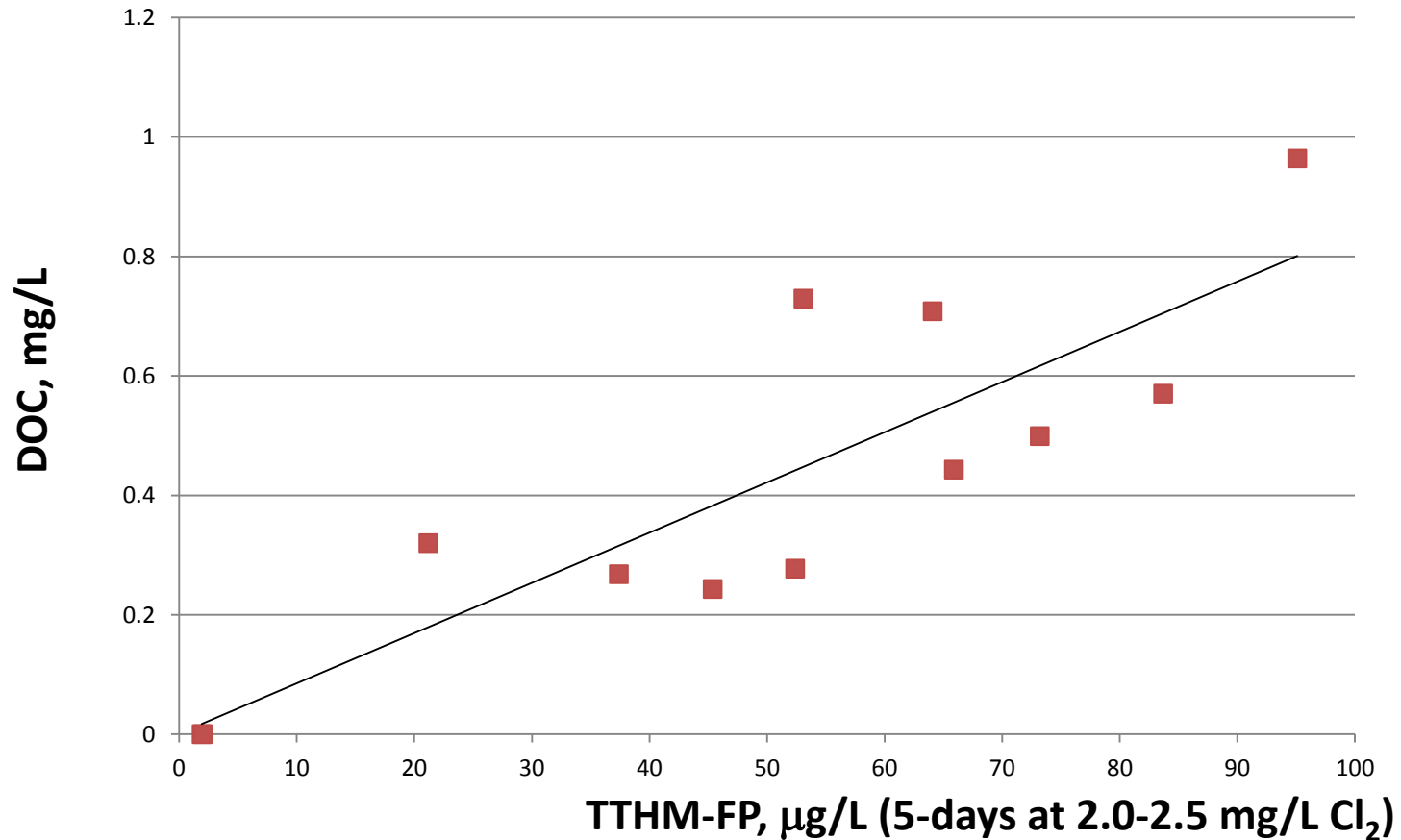
# Mountains of Data



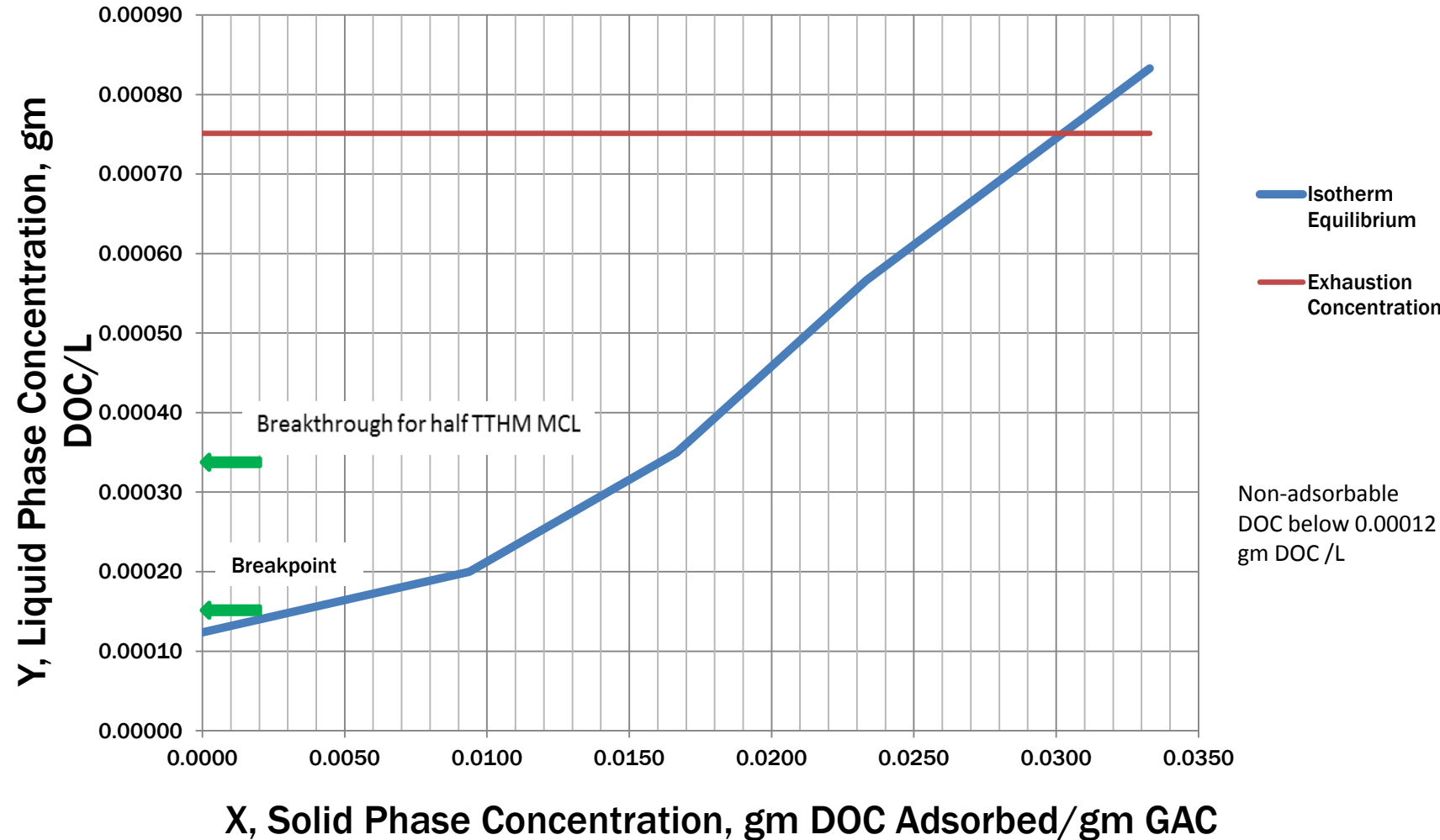
# Isotherm Testing

- Performed by Calgon
- Correlated TTHM-FP with TOC and DOC
- Batch Testing of Raw Water with Different Concentrations of Pulverized GAC
  - Filtrasorb 300 (bituminous)
  - Filtrasorb 400 (bituminous)
  - OLC (coconut shell)
- Filtrasorb 400 Proved Best

# DOC Correlation with TTHM-FP



# DOC Isotherm Equilibrium Graph Filtrisorb 400



# Calgon Model 10 Adsorber



**10 Foot Diameter**

**20,000 Pounds**

**Filtrisorb 400 per Tank**

**Operated at 0.5 MGD**





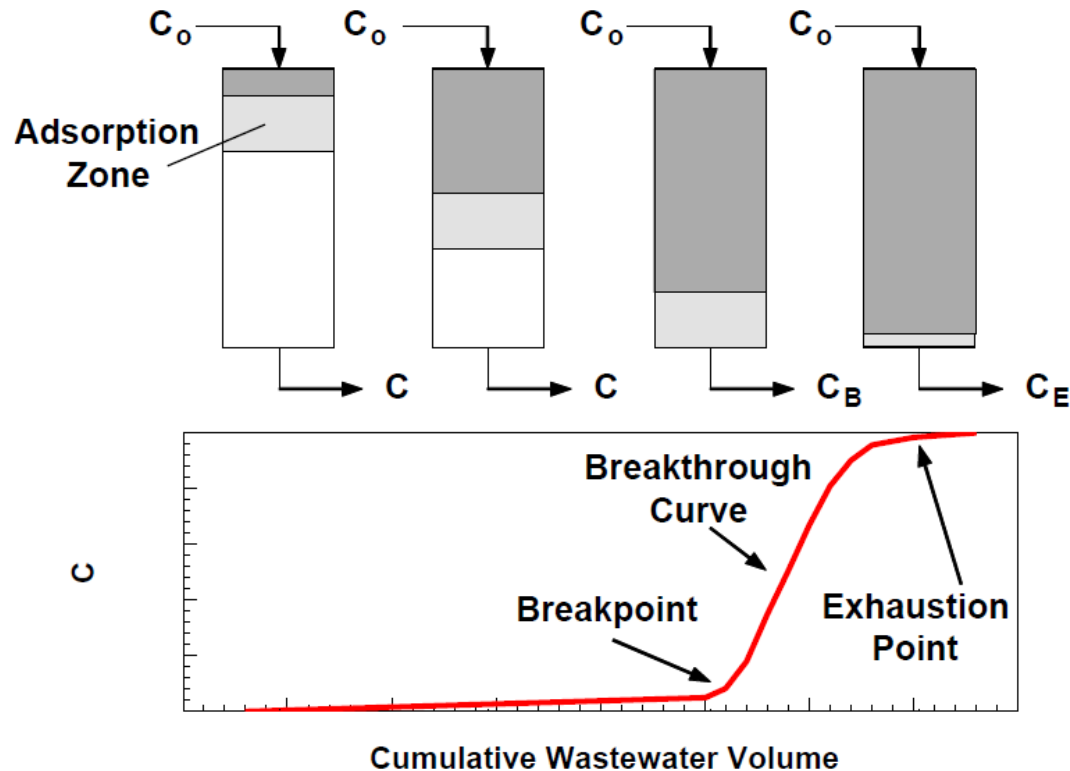
# Typical Analysis



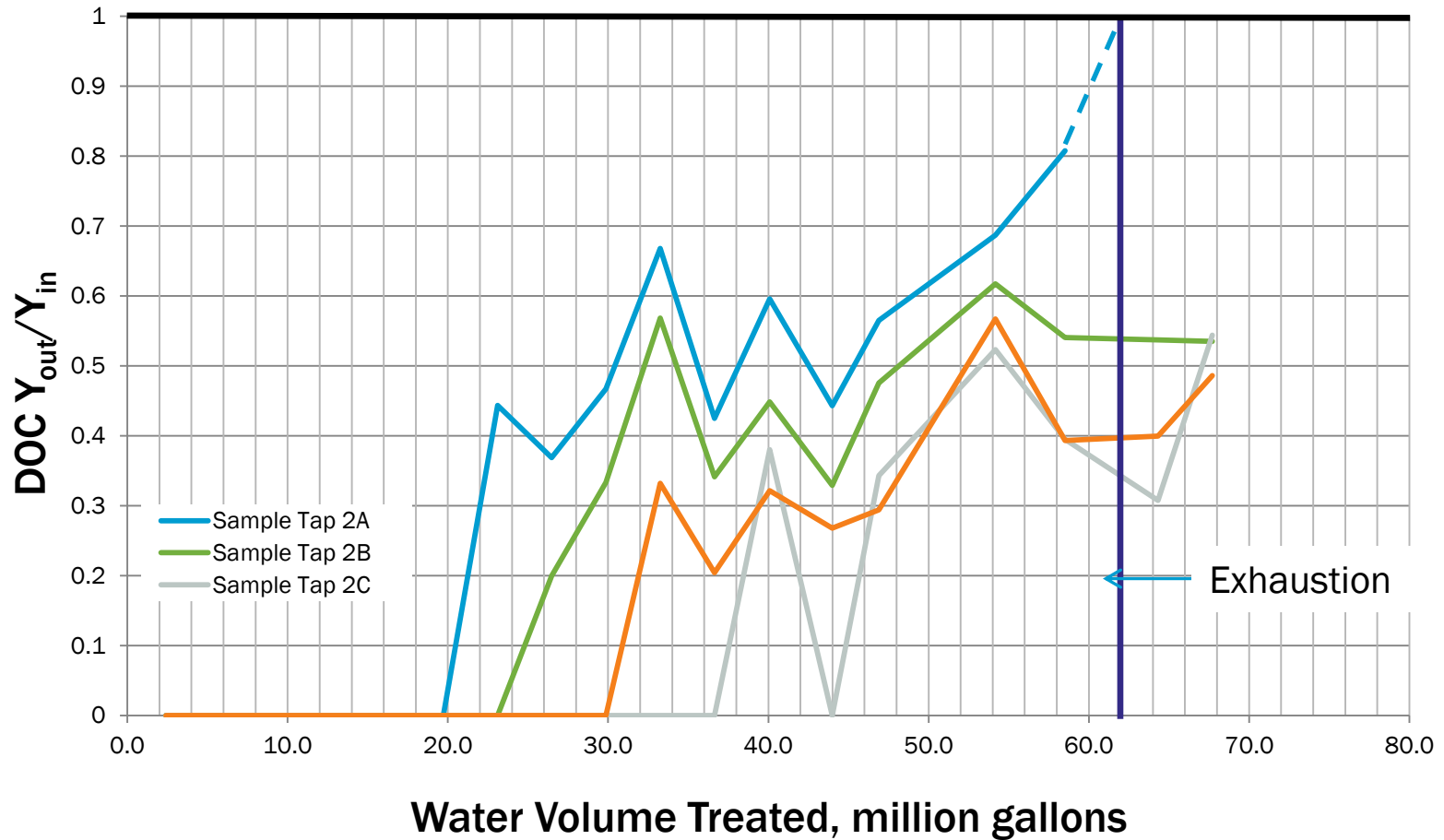


# Typical Concentration Profile Across Downward Flow GAC Adsorber

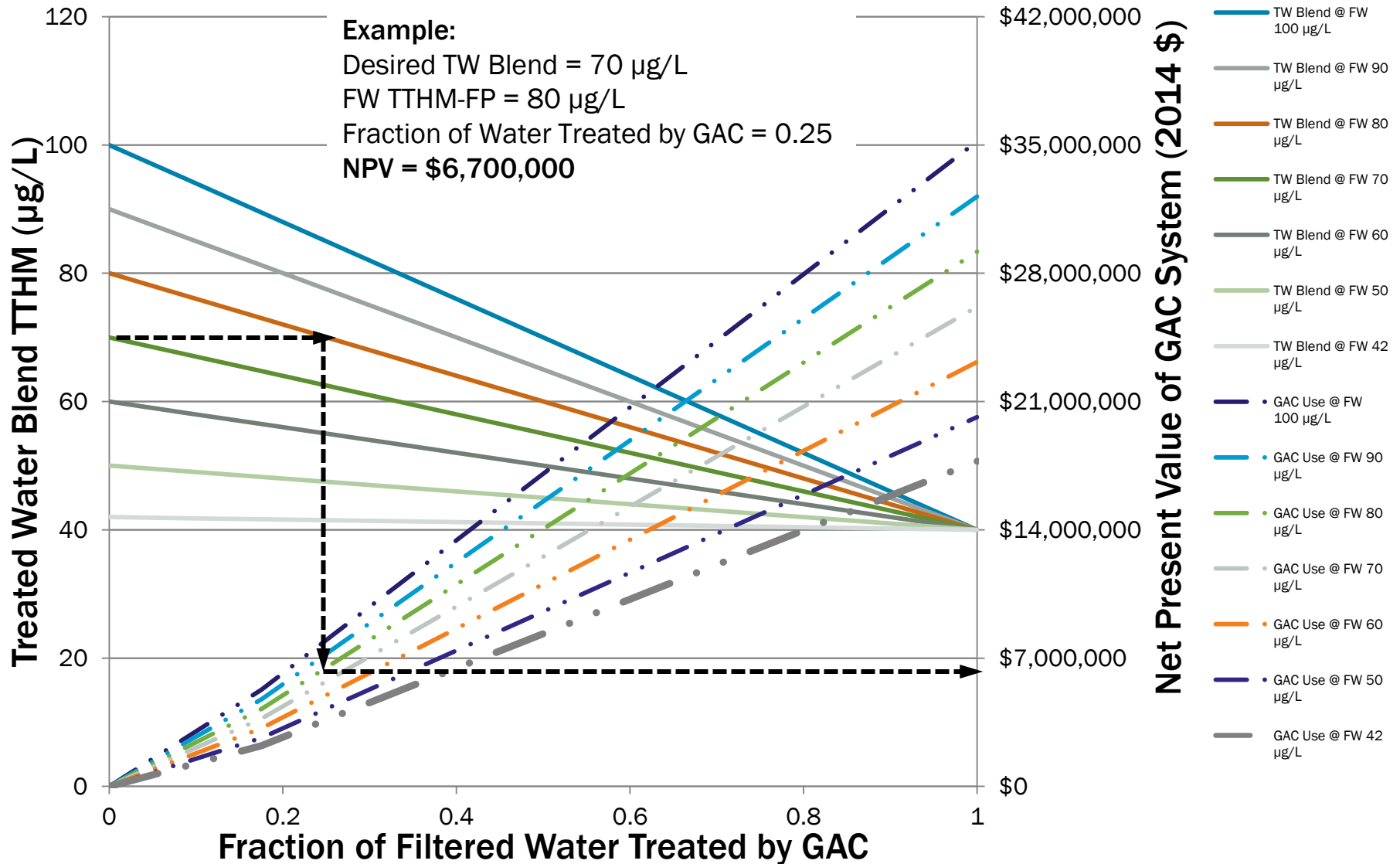
## Breakpoint and Breakthrough Curve



# Adsorber 2 DOC Removal and Volume Treated



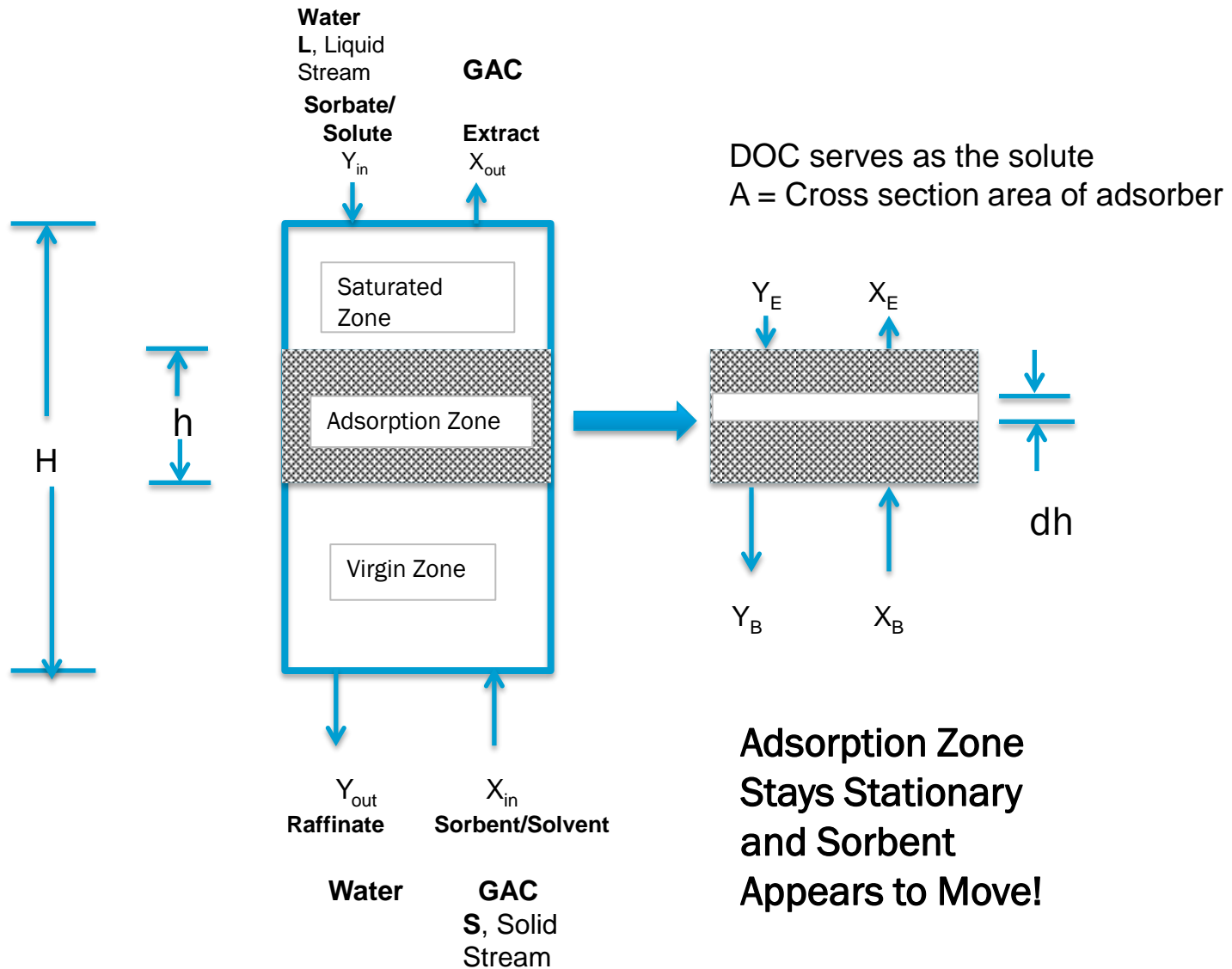
# Composite Blending Nomograph



# Expanded Analysis



# Counter Current Mass Balance Model

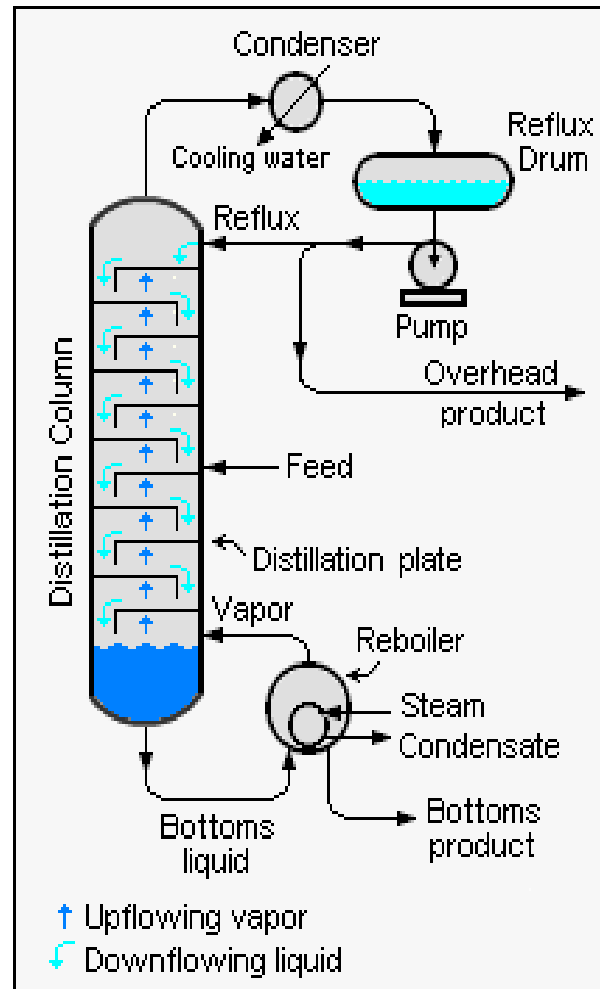


# GAC Adsorber Similar to Distillation Tower

## Continuous Mass Transfer Instead of Discrete Steps

**NTU:**  
Number  
of plates

**HTU:**  
Height  
of each  
plate





# Mass Balance Analysis-1

- $LY_{in} + SX_{in} = LY_{out} + SX_{out}$
- $S/L$  = slope of operating line, gm GAC/L feed water
- $(L/A)dY/dh = (S/A)dX/dh = K_L a(Y_{ops} - Y_{eq})$ 
  - $K_L a$  – volumetric mass transfer coefficient based on liquid phase driving forces
  - $Y_{eq}$  – equilibrium DOC concentration
- Perform Integration
$$h = L/AK_L a \int dY/(Y_{ops} - Y_{eq}), \text{ integrate from } Y_B \text{ to } Y_E$$
$$\text{NTU, number of transfer units} = \int dY/(Y_{ops} - Y_{eq}), \text{ integrate from } Y_B \text{ to } Y_E$$
$$h = (L/AK_L a)\text{NTU}$$
- **NTU: Solve graphically (McCabe-Thiele method) or by numerical integration**

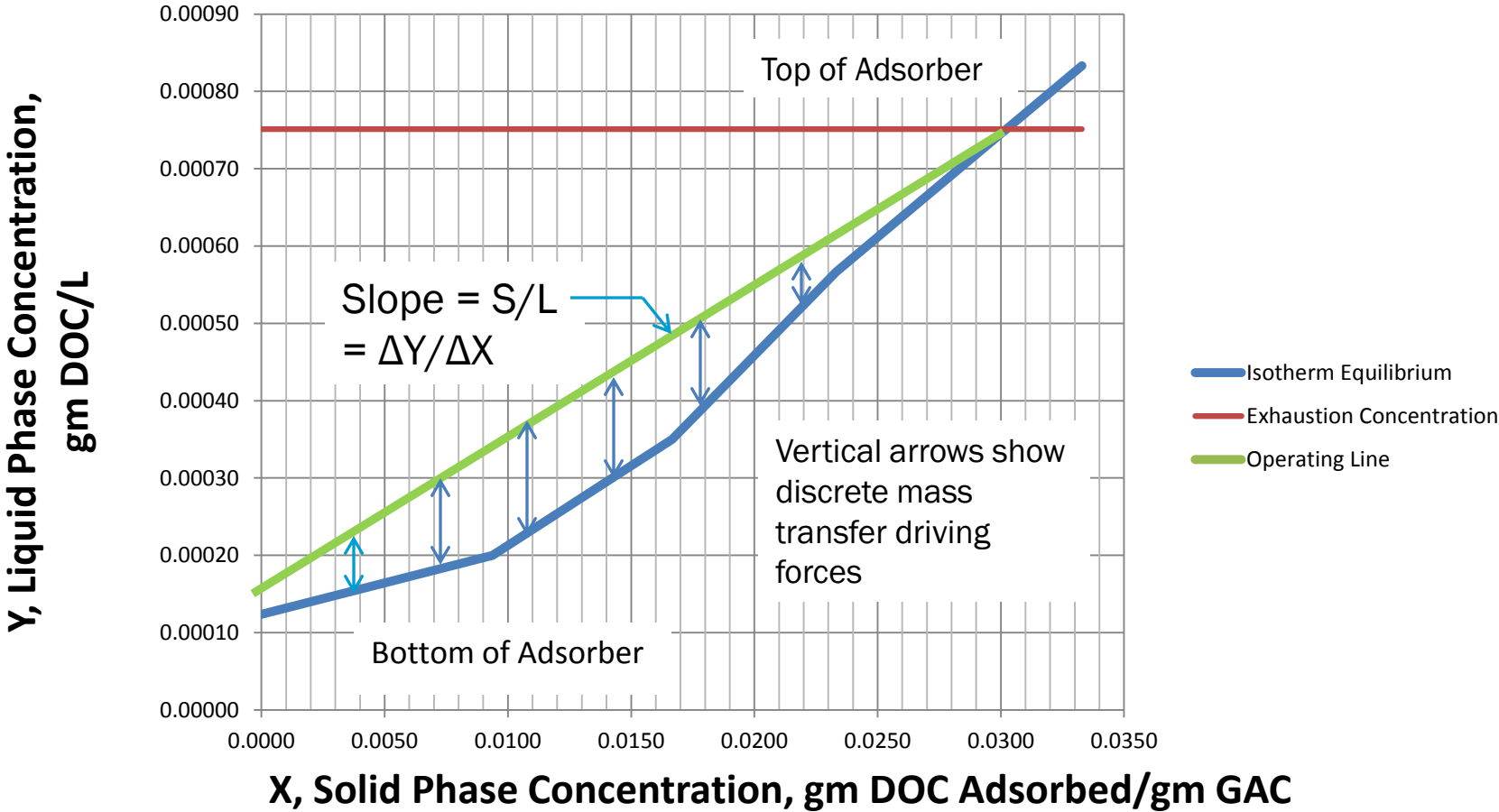
# Mass Balance Analysis-2

- HETP (HTU) =  $L/AK_La = U_L/K_La$       $U_L$  – liquid phase velocity
- $K_La$ : Liquid phase volumetric mass transfer coefficient
- $K_La = \alpha U_L^{0.5}$ 
  - Applicable for this type of aqueous solution mass transfer
  - For this case,  $\alpha$  is a constant
  - Diffusivity and particle size is constant
- $h = \text{HTU} \times \text{NTU} = \text{height of adsorption zone}$

# $K_L a$ is Important

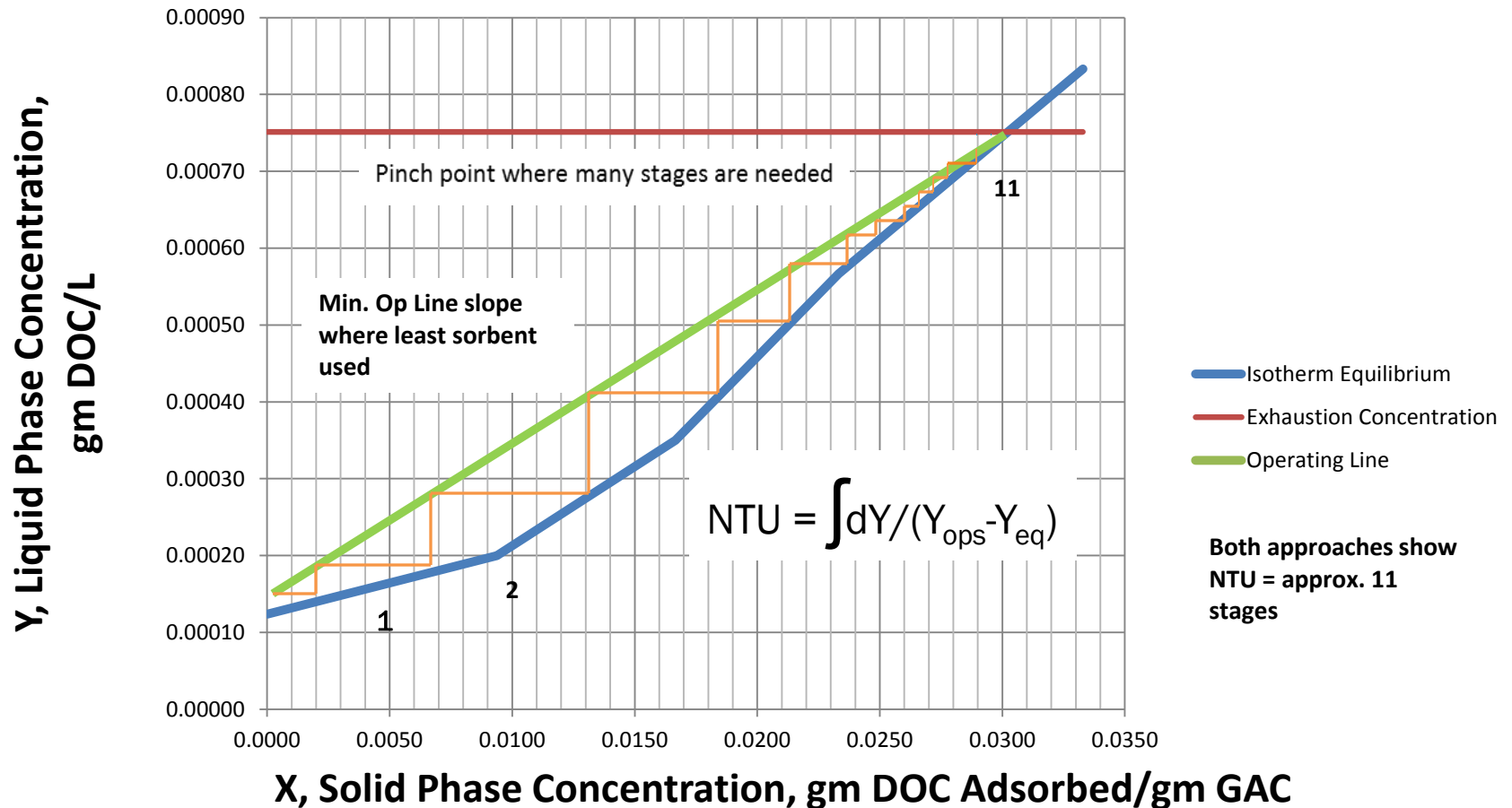
- Determined experimentally
- Useful for understanding mass transfer
- Underlying assumptions
  - Liquid phase mass transfer step is predominant
  - Other mass transfer resistances (diffusion and adsorption) are less significant
  - No chemical reaction (e.g. oxidation)
- Can investigate adsorber geometries and loading

# Equilibrium and Operating Lines



# 2 Ways to Determine NTUs

## McCabe-Thiele Graph or Numerical Integration



# Other Results of Mass Balance Analysis

NTU = 11.1 stages

HETP (HTU) = 14.3 cm

h, height of adsorber zone = 158 cm

$U_L = 17.4$  cm/min.

**$K_L a = 1.22$  min<sup>-1</sup>**

$K_L a = \alpha U_L^{0.5}$

**$\alpha =$  Experimentally calculated =  $0.29$  (cm-min)<sup>-0.5</sup>**

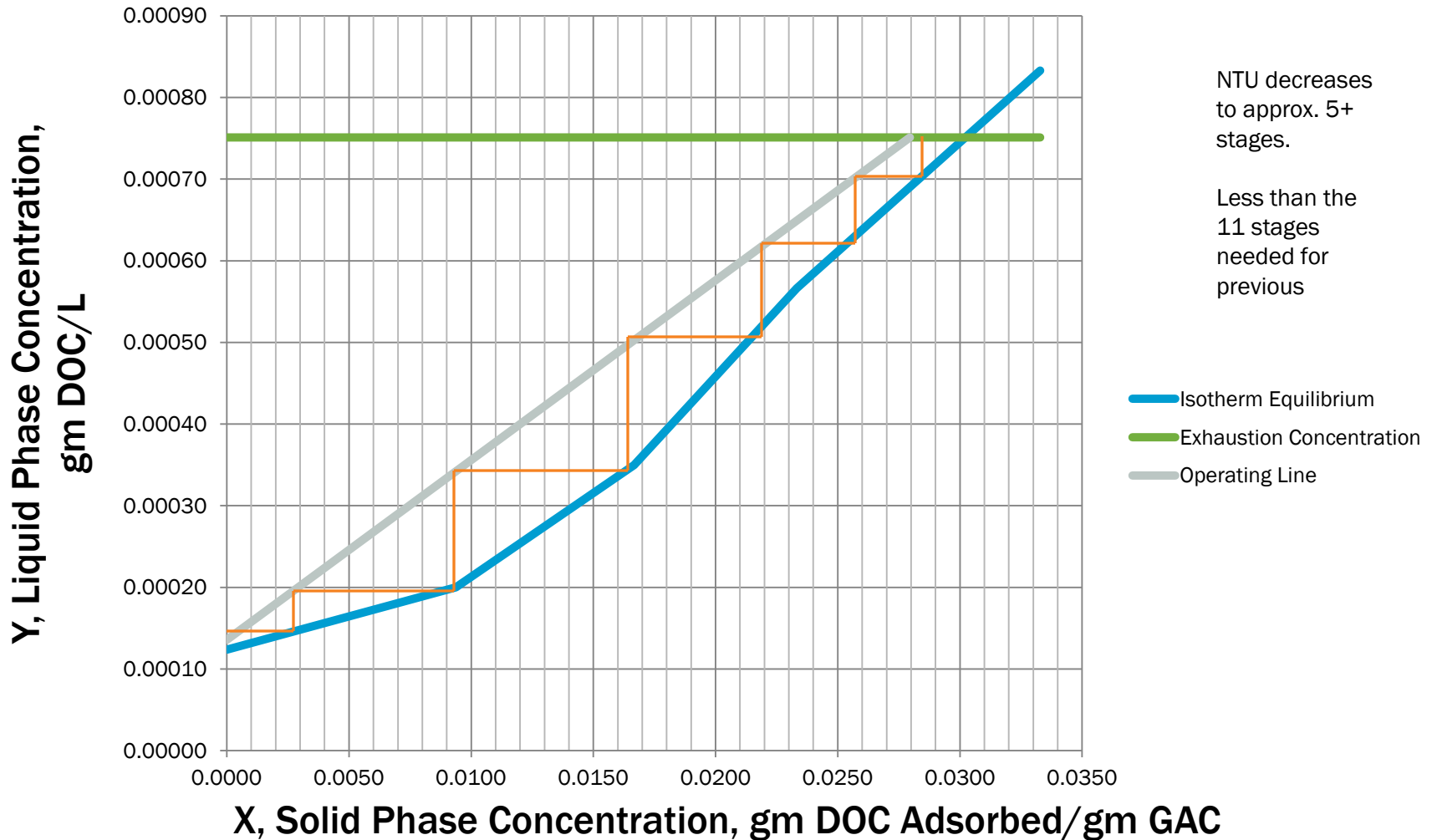


# Broad and Useful Interpretations

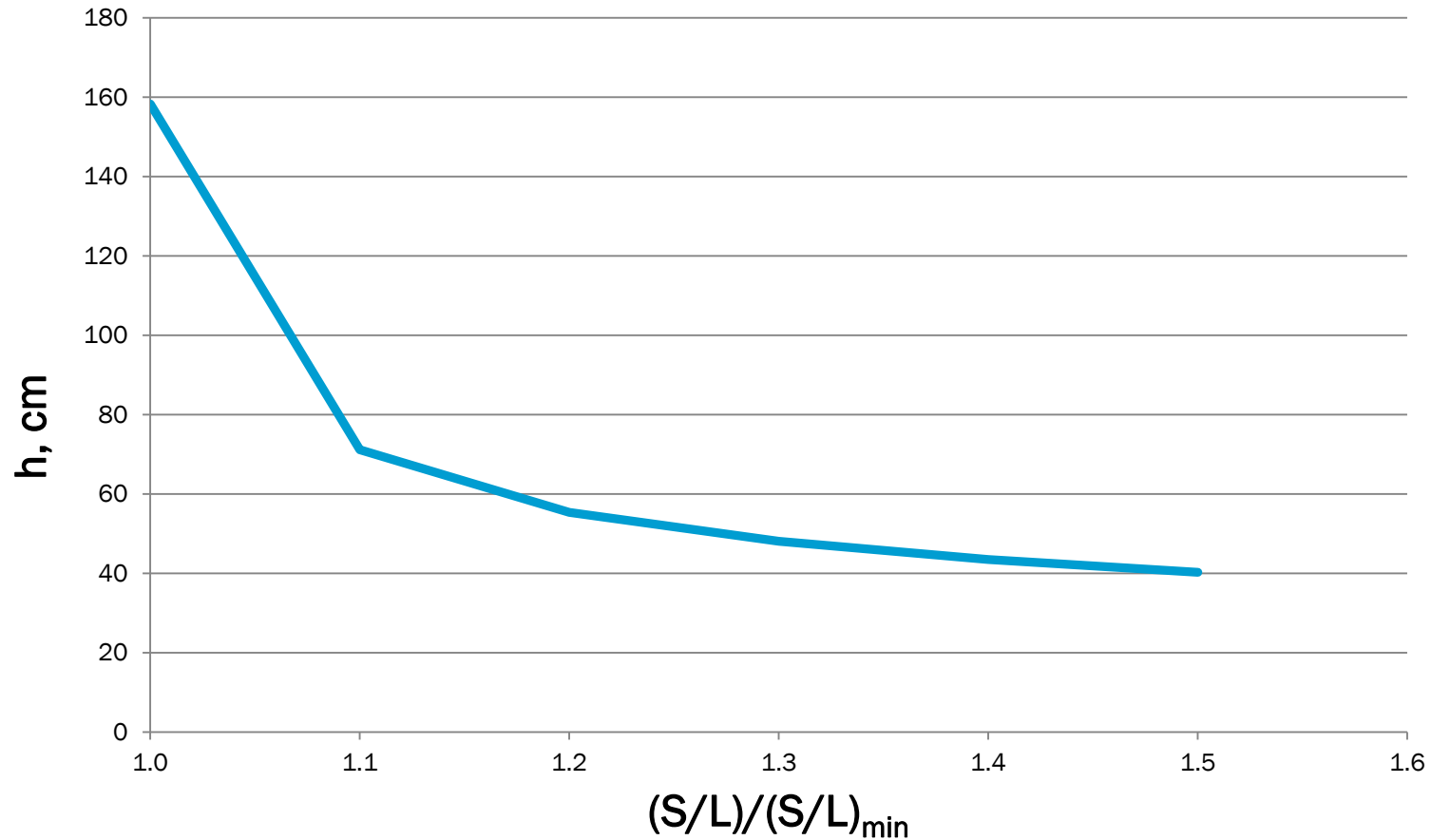


# $K_L a$ Data Allows Investigation of Other Operating Schemes for Design

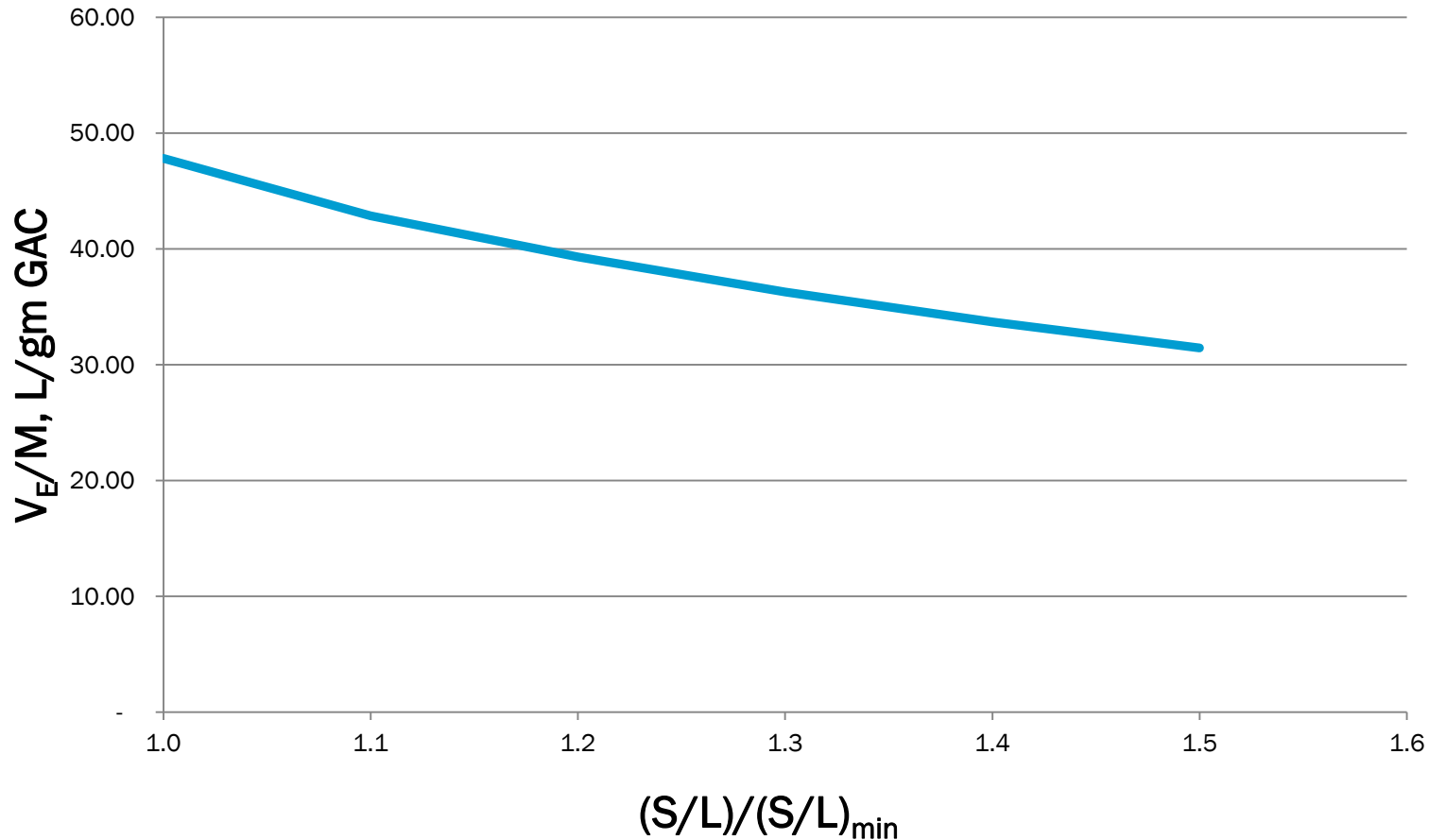
Example: Increase Op. Line Slope by 10%



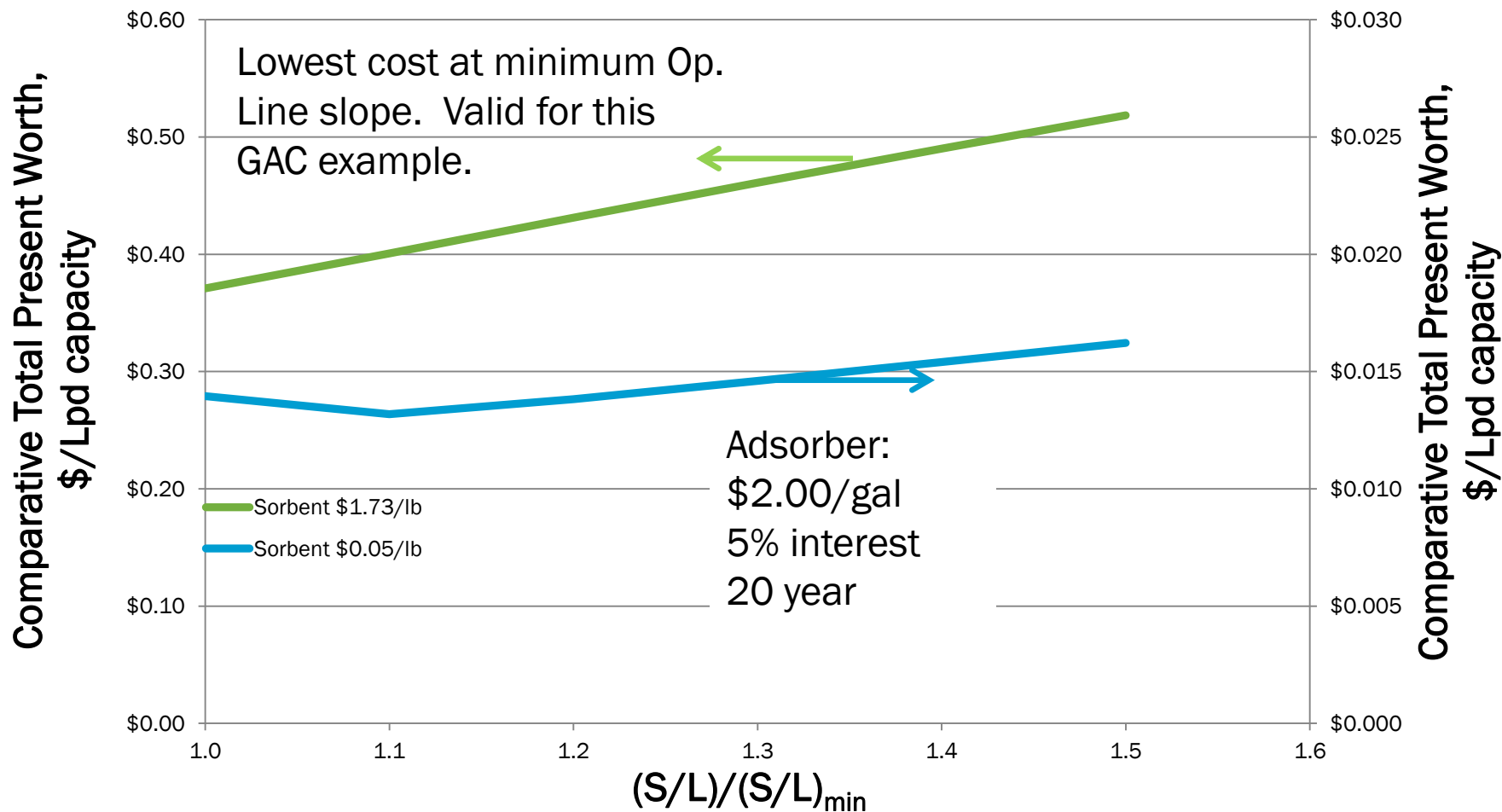
# Influence of Operating Line Slope to Min. Slope on $h$ , height of Adsorber Zone



# Influence of Operating Line Slope Ratio to Raffinate Volume per Unit Mass of Sorbent Consumed



# Economic Analysis of Differing Operating Line Slopes

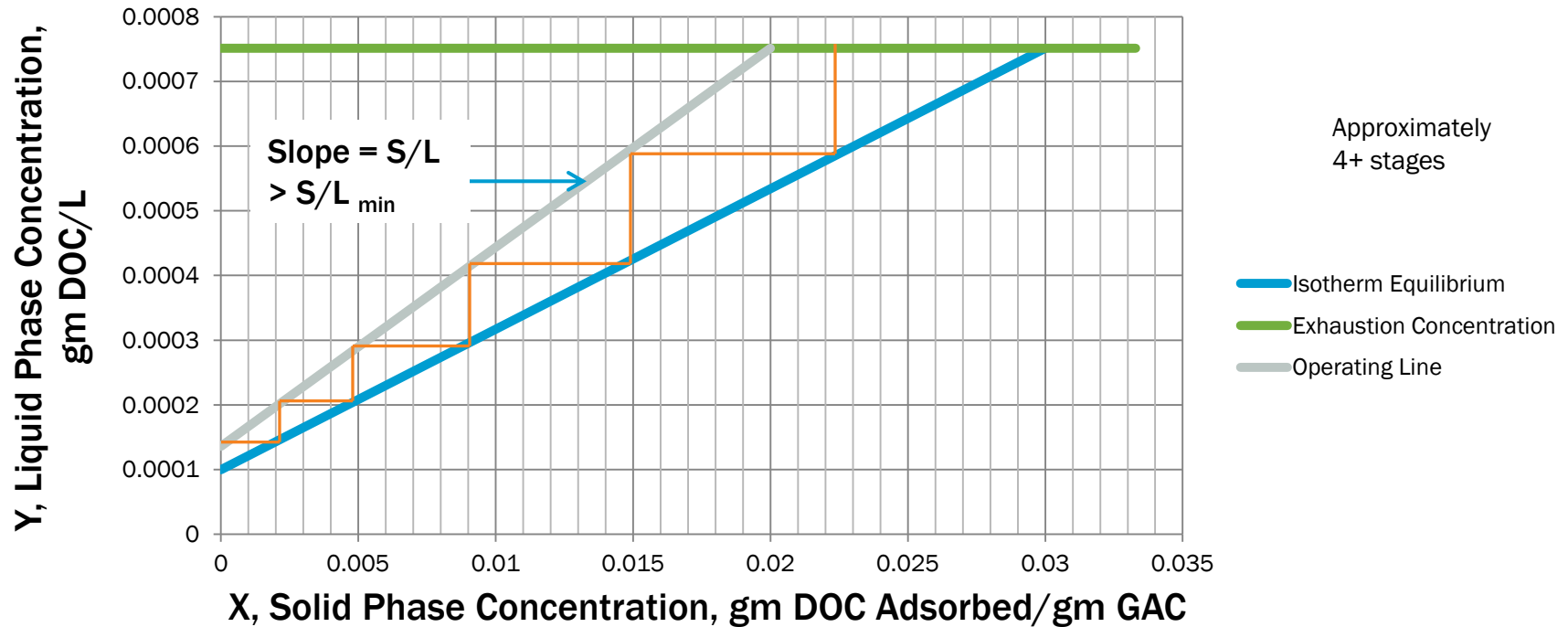


# Use of Experimental $K_L$ a Determination on Design Considerations

- Permits assessment of different design parameters
  - Adsorber diameters
  - Liquid stream flow velocity
  - Number of stages
  - Adsorber height
  - Different sorbents (of similar particle size)
  - Different water quality targets
- Search for Optimal Adsorber Configuration
- **For Costly Sorbents (e.g. GAC),  $S/L_{\min}$  is justified**
- For inexpensive sorbents/solvents or expensive tank construction, optimal  $S/L$ -may be larger than  $S/L_{\min}$  (perhaps 1.2 to 1.4)
- Approach can be used for other related mass transfer systems
  - Ion exchange
  - Gas adsorption/stripping
- **Adjust Model and Account for Biological Oxidation!**

# What to do if Results are Poor

- Less Than Favorable (Flat) Equilibrium-Poor Driving Force



- Try new sorbent
- Assess other approaches

# Summary

- Mass Balance Model is Useful
- Benefits of Evaluating  $K_L a$ 
  - Permits Varied Design Assessments
  - Determines Optimal Design
  - Determines Optimal Operational Strategies



# Questions and Discussion

