

Manganese Sequestration in Potable Water – A Quantitative Evaluation

Michael Britton, P.E.

(509) 546-2074, Michael.Britton@hdrinc.com

Glen Boyd, PhD, P.E., Andrew Hill, P.E.,

Alexander Vetrovs, Steve Reiber, PhD,

HDR Engineering, Inc.

Gregory Keith, City of Issaquah

Outline

- Information on Manganese, Polyphosphates, and Sequestration
- Water Quality
- Sequestrants
- Test Method
- Results
- Conclusions

Manganese Information

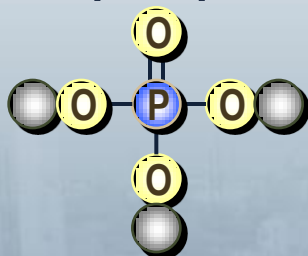
- No Primary MCL
- Secondary MCL – 0.050 mg/L
 - Aesthetics - color and staining
- Typical Treatment Target
 - 0.010 to 0.020 mg/L



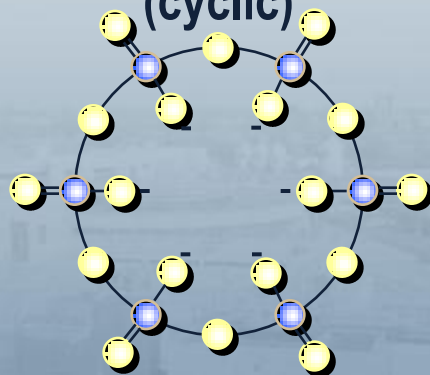
The Family of Phosphates

- Most polyphosphate sequestrants are really a soup consisting of all these compounds

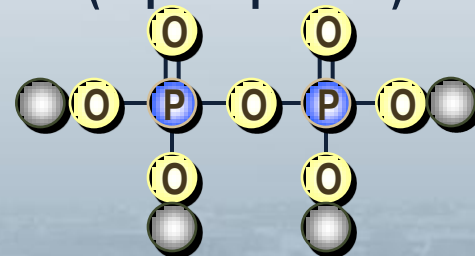
Orthophosphates
(Monophosphates)



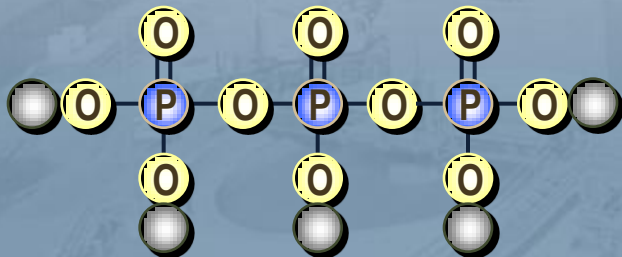
Metaphosphates
(cyclic)



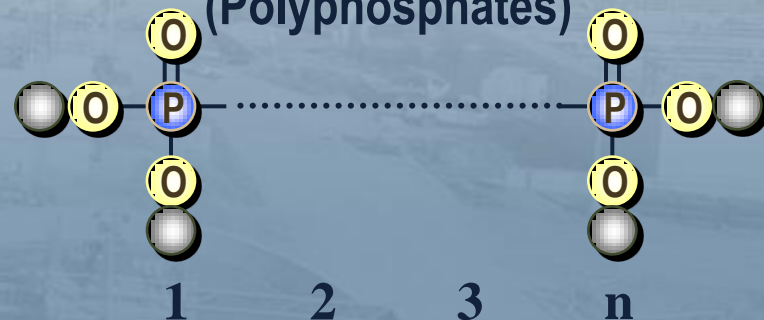
Pyrophosphates
(Diphosphates)



Tripolyphosphates

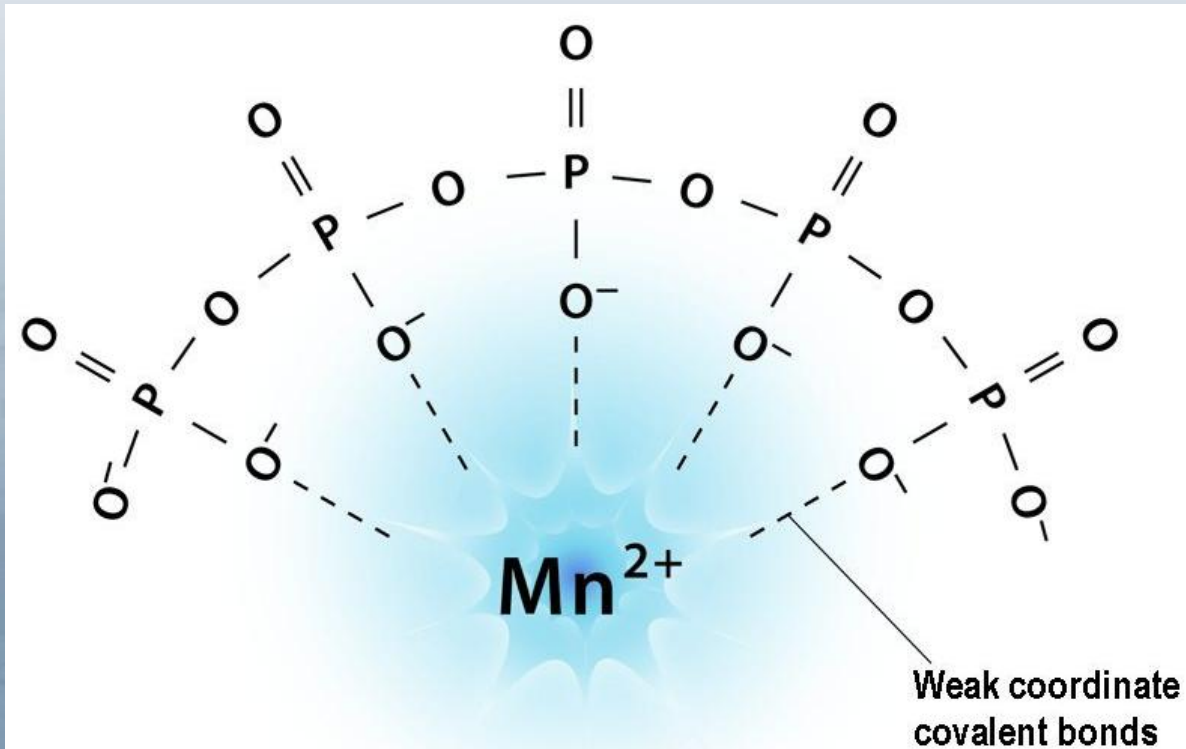


Condensed phosphates
(Polyphosphates)



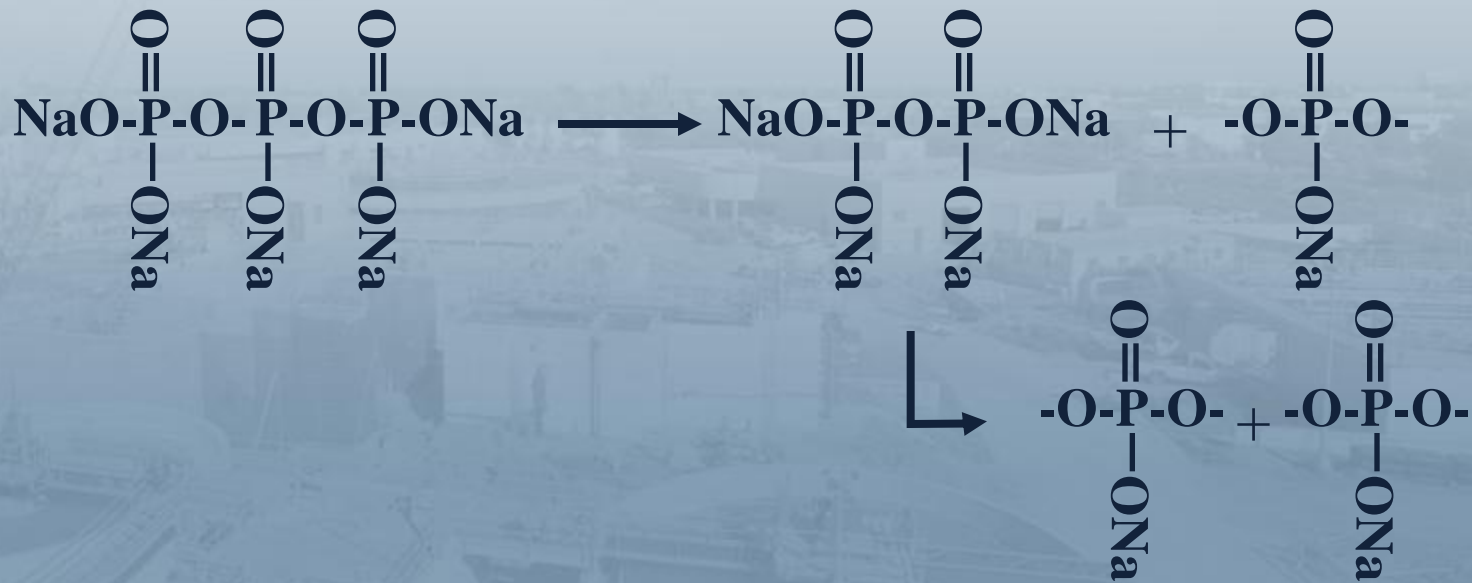
Theoretical Poly-P / Manganese Complex

- Parameters Affecting Complex Stability
 - pH
 - Temperature
 - Poly-P chain length



The Problem with Poly-P: Hydrolysis

- Slow reaction with water to yield orthophosphate – loss of complexation capability
- Rate of hydrolysis is related to temperature and pH
- Time release – Controllable kinetics based on polymer length
- Longer chain polyphosphates retain sequestration efficiency longer



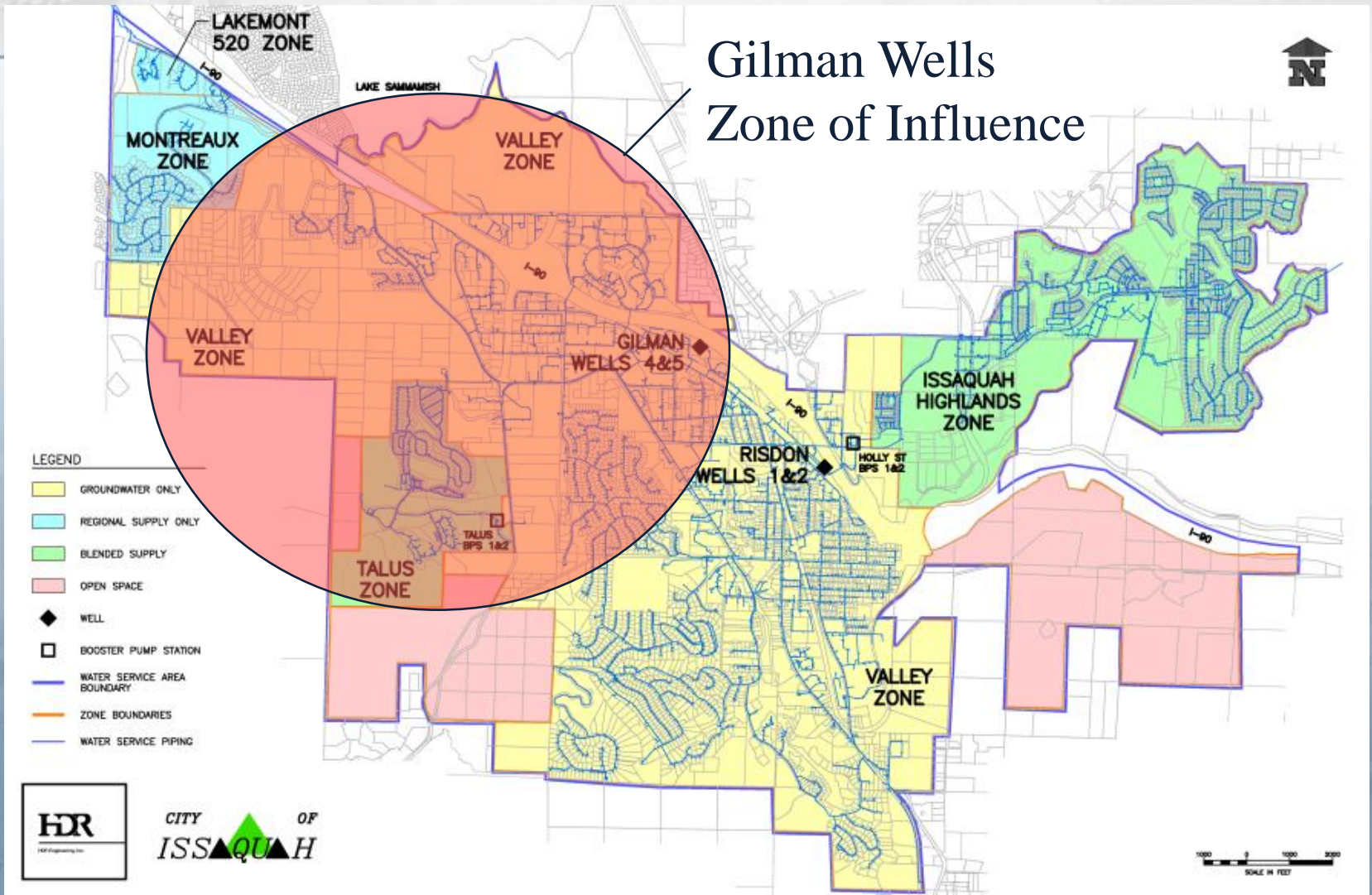
Parameters Affecting Sequestrant Efficiency

- pH
- Temperature
- Poly-P chain length
- Sequestered water age
- Chlorine concentration
- Mn concentration
- Completing ion concentration (i.e., TDS, hardness)

City of Issaquah, Washington State



City of Issaquah Water Distribution System



Water Quality

| Parameter | Gilman 5 | Gilman 4&5 | Risdon 1&2 |
|---|---------------------|---------------------------|---------------------------|
| Manganese, Total (mg/L) | 0.090 | 0.078 | ND |
| Iron, Total (mg/L) | ND | 0.034 | ND |
| Calcium (mg/L) | 34 | 32 | 16 |
| pH | 8.4 | 7.8 | 8.0 |
| Alkalinity (mg/L as CaCO ₃) | 103 | 102 | 67 |

ND: Non-Detect

Sequestrant Information

| Seq. | Strength (% PO₄) | S.G. (mg/L) | Recommended Dose (mg/L as PO₄) | NSF 60 Certified |
|-------------|--|------------------------|--|-----------------------------|
| A | 35% | 1.73 | 1.00 | Yes |
| B | 29% | 1.32 | 1.45 | Yes |
| C | 35% | 1.40 | 0.35 | Yes |
| D | 33% | 1.40 | 0.50 | Yes |
| E | 25% | 1.51 | 1.00 | Yes |
| F | 23% | 1.29 | 1.00 | Not Found |

Test Matrix

| Round | Water | # of Seq. | Seq. Dose (mg/L) |
|-------|---|-----------|------------------|
| 1 | Well 5 | 6 | 1X |
| 2 | Gilman (Well 5+4) | 6 | 3X |
| 3 | Gilman (Well 5+4) | 2 | 1X, 2X, 3X, & 5X |
| 4 | Gilman:Risdon 100:0, 90:10, 50:50, & 10:90 | 1 | 5X |

Test Procedure - General

- Chlorine burn out of amber glass bottles
 - Soaked for ≥ 24 hours with 3 to 3.5 mg/L chlorine
 - Rinsed 3 times with DI water
- Each test included a control
 - Test water with no sequestrant
- Chlorine doses target
 - 0.5 mg/L Cl_2 Residual after 48 hours
- Constant Temperature
 - $60 \pm 5^\circ\text{F}$

Test Procedure

Test
Water



Seq.



Hypo



Stir
30-60
sec.



Stir
10 sec.



Round 4
Only

Rison
Water



Stir
2 min.



Filter
0.45µm
Free (Mn)



Total and
Free Mn
Analysis



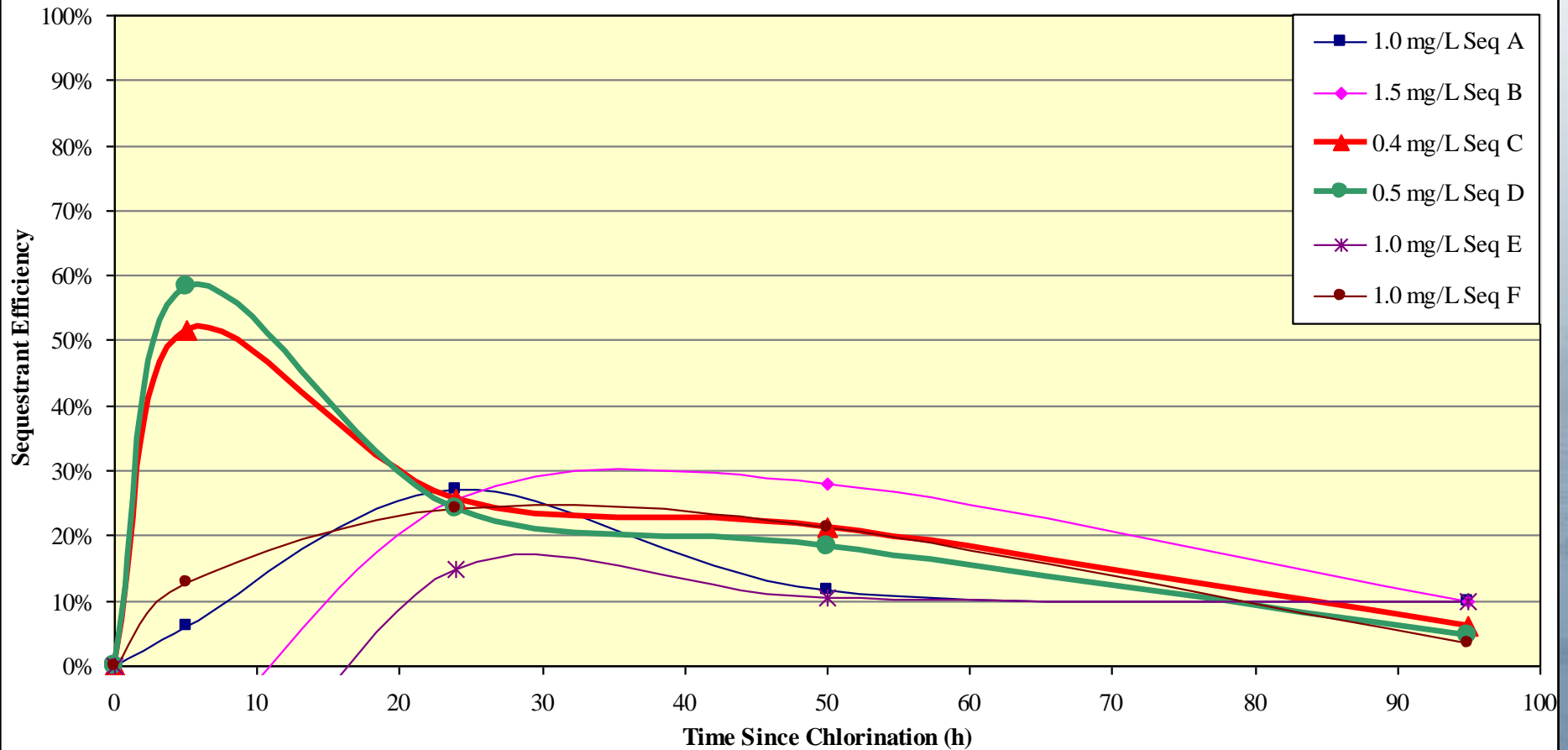
Calculations

- Soluble Mn
 - Mn that passes through 0.45 µm filter
- Precipitated Mn @ time t
 - Soluble Mn @ time 0 – Soluble Mn @ time t
- Sequestrant Efficiency
 - Percent reduction in Mn precipitation due to sequestrant

$$\frac{\text{Prec. Mn Control} - \text{Prec. Mn Test}}{\text{Prec. Mn Control}} * 100\%$$

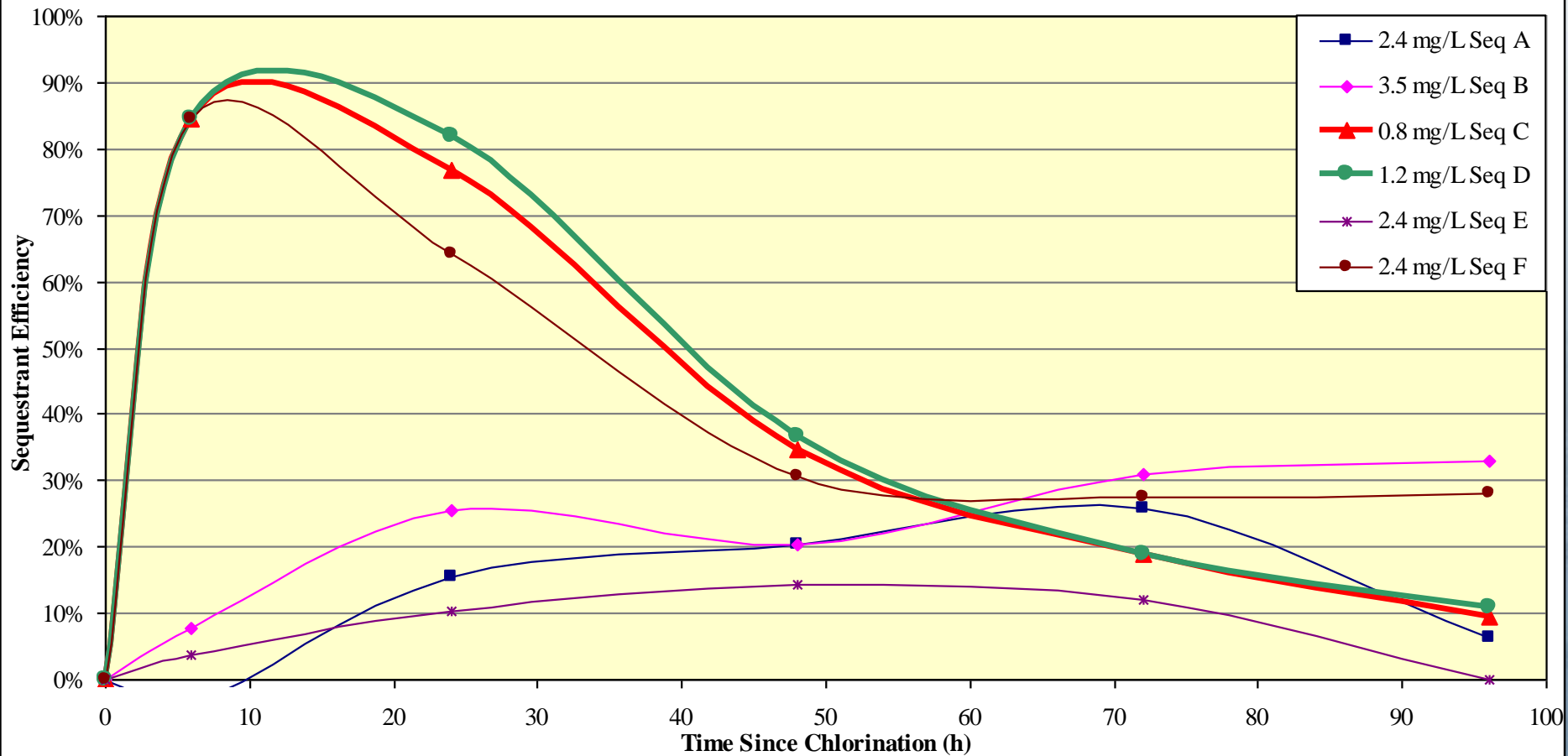
Round 1 – Sequestrant Efficiency

Well 5 Water @ 60°F & 1.5 mg/L Cl₂
Manufacturer's Recommended Sequestrant Dose



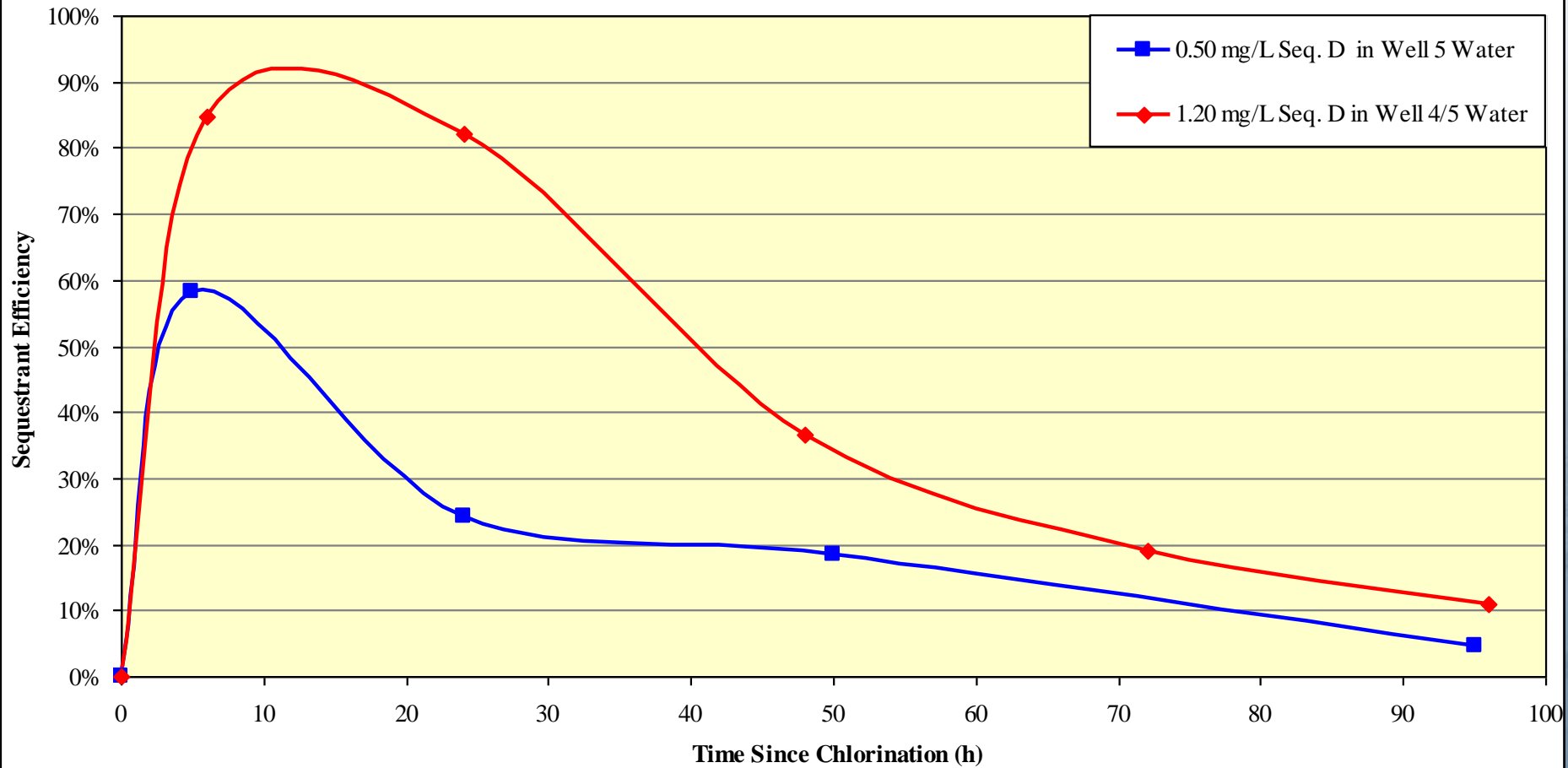
Round 2 – Sequestrant Efficiency

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂
3X Manufacturer's Recommended Sequestrant Dose



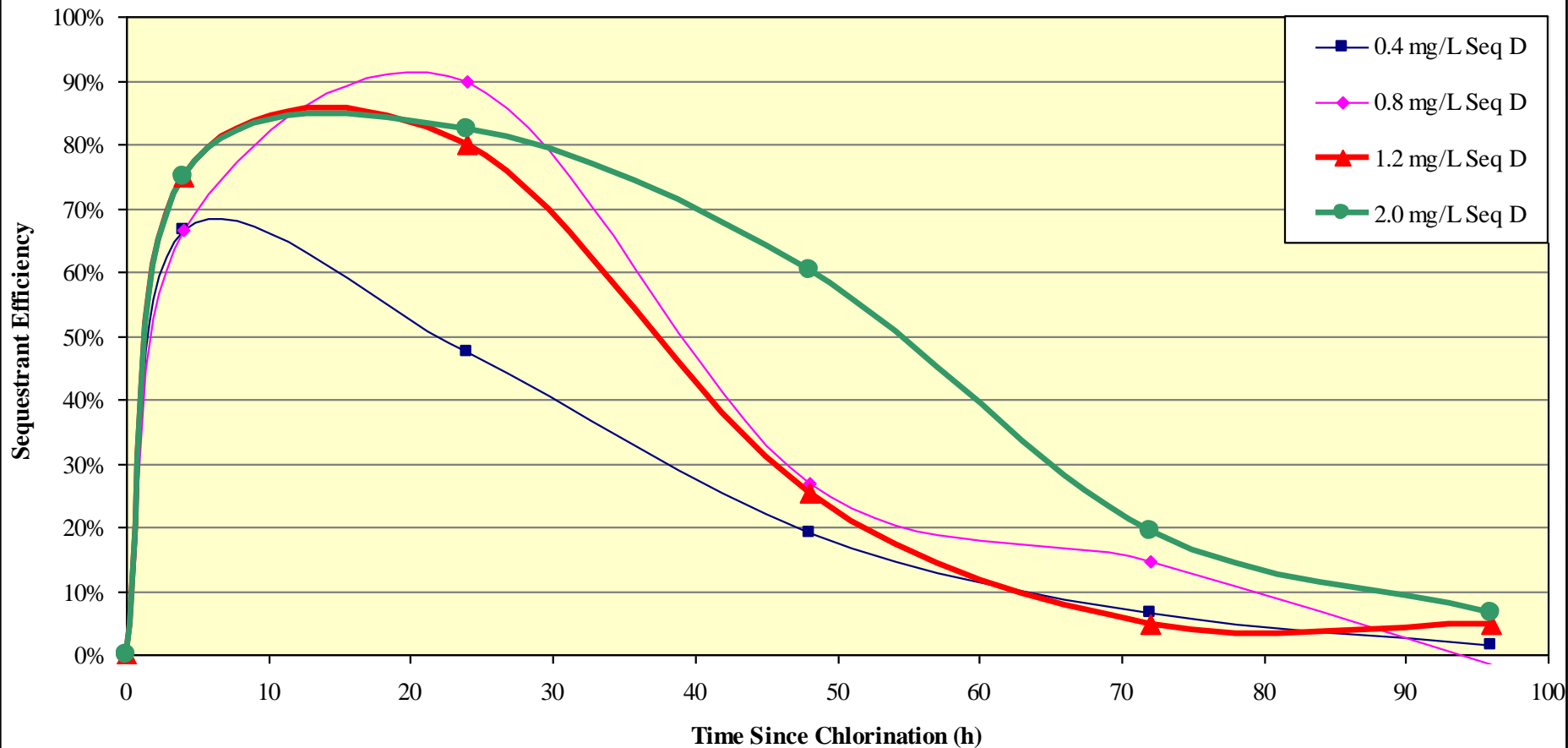
Comparison of Round 1 and 2 – Seq. D

**Seq. D Sequestrant Efficiency
in Well 5 Water vs. Well 4/5 Water**



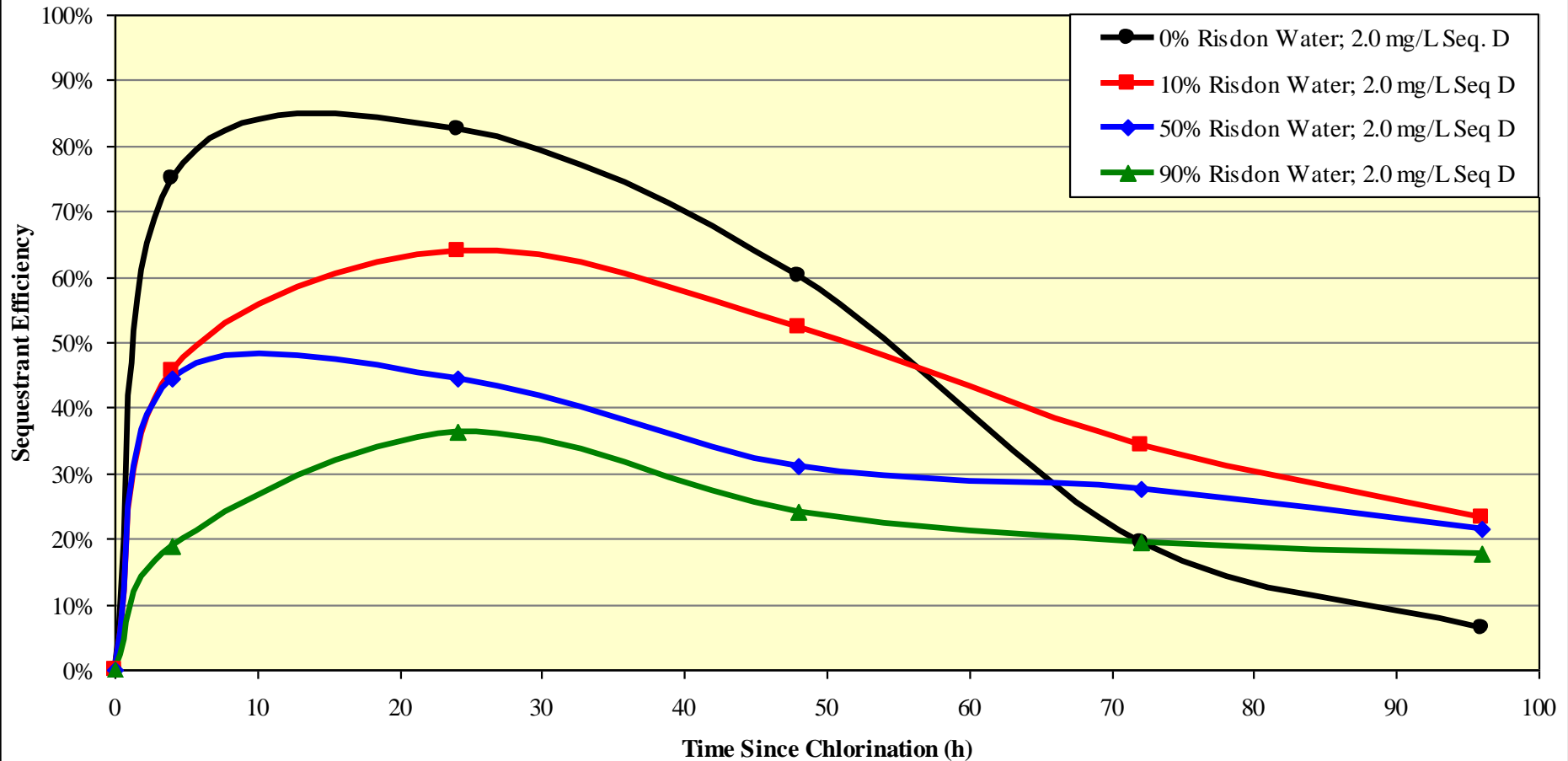
Round 3, Seq. D – Sequestrant Efficiency

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂
Sequestrant D @ 1, 2, 3, & 5 X Manufacturer's Recommended Dose



Round 4 – Gilman/Risdon Blended Water

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂ Blended with Risdon Well Water
2 mg/L Sequestrant D in Gilman Water

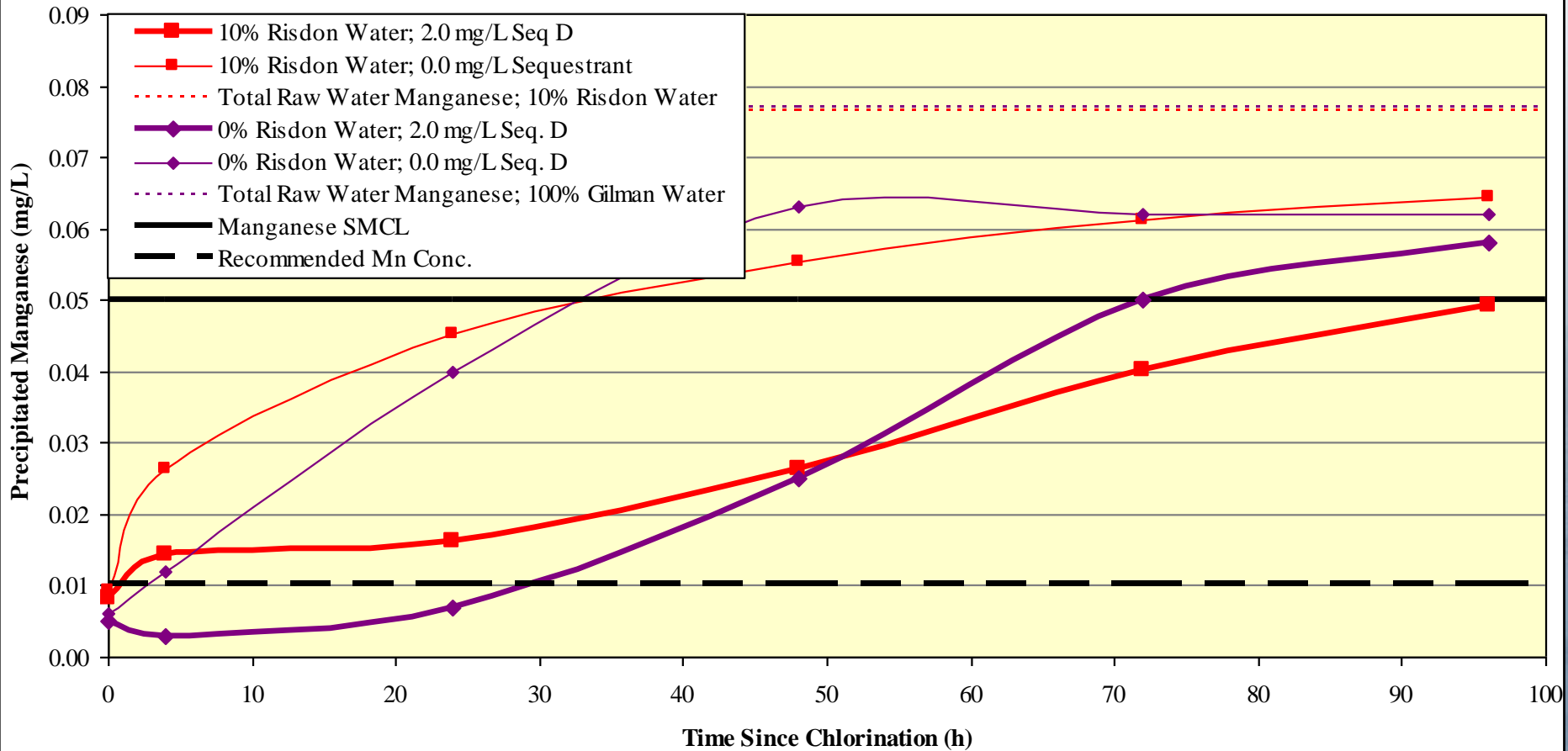


Diluting Mn Requires New Graph

- Sequestrant Efficiency
 - = % reduction in Mn precipitation due to sequestrant.
- Blending
 - Reduces original Mn
 - Reduces Mn precipitation.
- For blending tests, look at total Mn precipitation, not Sequestrant Efficiency

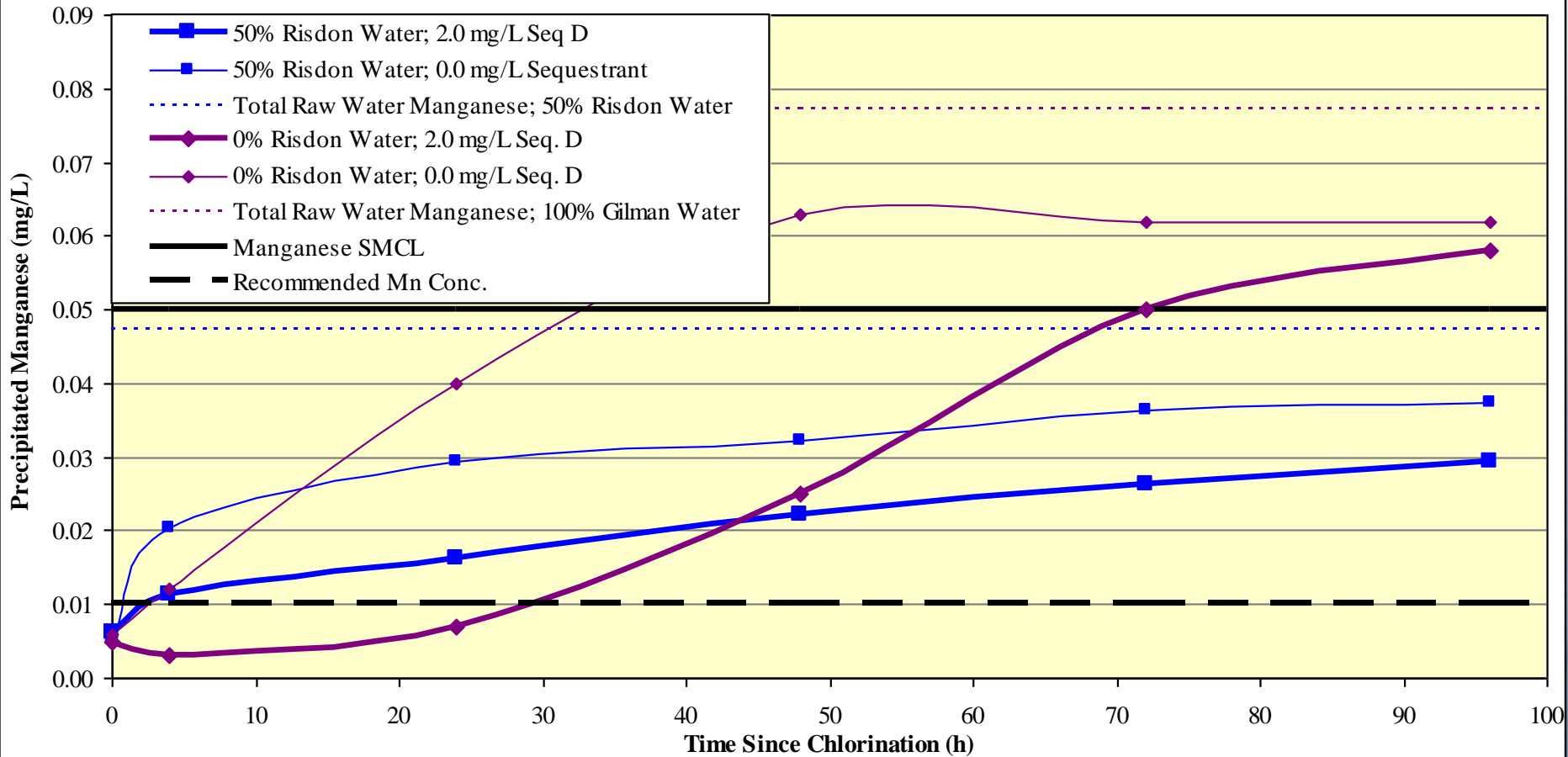
Round 4 – 90:10 Gilman to Risdon Water

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂ Blended with 10% Risdon Well Water
2 mg/L Sequestrant D in Gilman Water



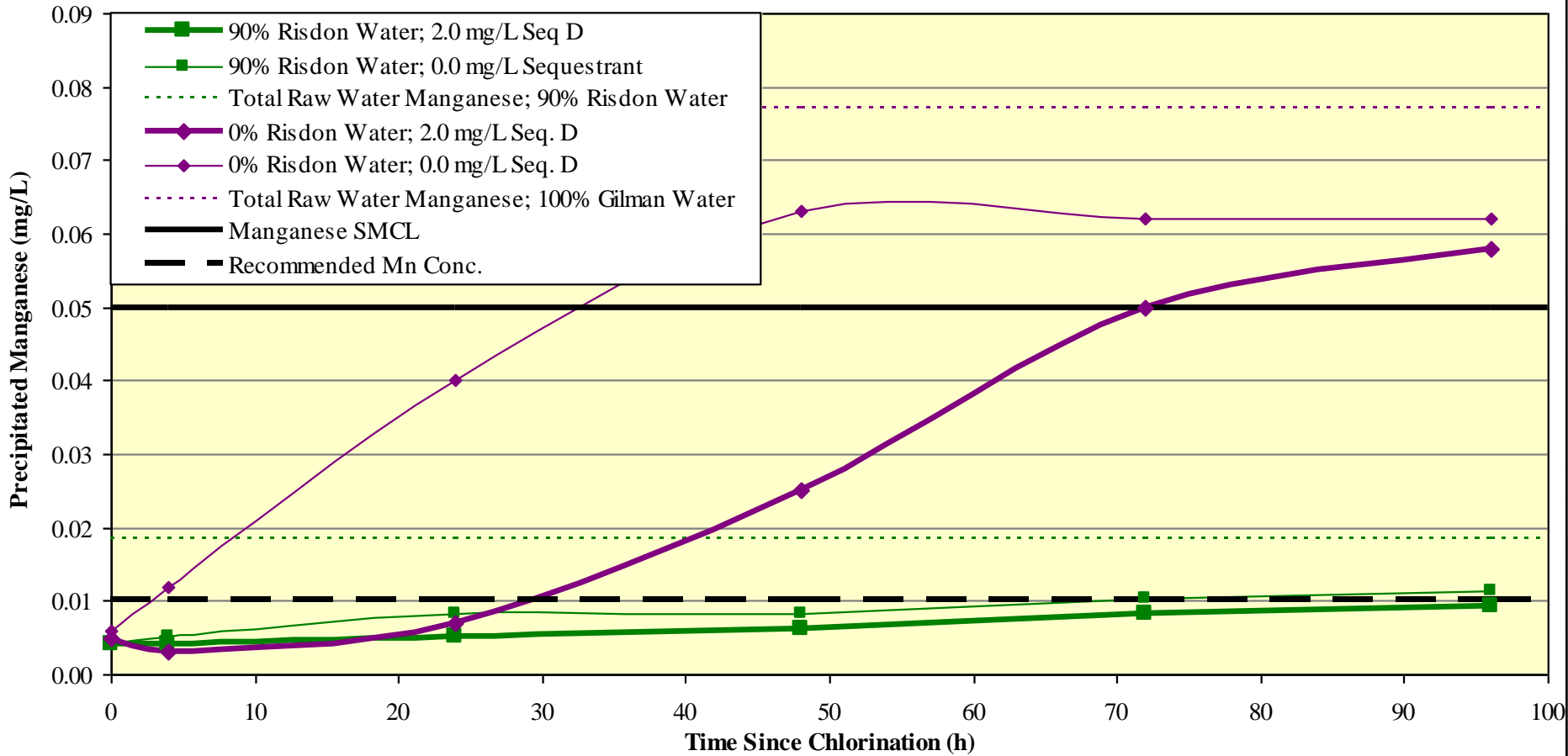
Round 4 – 50:50 Gilman to Risdon Water

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂ Blended with 50% Risdon Well Water
2 mg/L Sequestrant D in Gilman Water



Round 4 – 10:90 Gilman to Risdon Water

Well 4/5 Water @ 60°F & 1.3 mg/L Cl₂ Blended with 90% Risdon Well Water
2 mg/L Sequestrant D in Gilman Water



Conclusions

- Demonstrated 2.0 mg/L Sequestrant D reduced manganese precipitation by:
 - 82% for 24 hours
 - 60% for 48 hours
 - 20% for 72 hours
- Sequestrants can reduce or delay manganese precipitation in chlorinated water, but not significantly longer than 72 hours
- Blending high Mn water with low Mn water reduces manganese precipitation more than the sequestrant

Conclusions

- Sequestrants do not completely eliminate Mn precipitation
- Sequestration reduces Mn precipitation problem by
 - Slowing Mn precipitation
 - Allowing Mn to precipitate over wider region
- This results in
 - Smaller settled precipitate layer in pipes
 - Less Mn stirred up in hydraulic disturbances

Conclusions

■ City of Issaquah

- Instituted sequestration at 3X recommended dose
- Plans to increase directional high-volume flushing
- Plans to increase dead-end flushing
- Plans to increase water reservoir cleaning
- Visually inspects flush water for Mn precipitates
- Anticipates 1 FTE for flushing

■ Results

- Online since July 2008
- No complaints

Manganese Sequestration in Potable Water – A Quantitative Evaluation

Michael Britton, P.E.

(509) 546-2074, Michael.Britton@hdrinc.com

Glen Boyd, PhD, P.E., Andrew Hill, P.E.,

Alexander Vetrovs, Steve Reiber, PhD,

HDR Engineering, Inc.

Gregory Keith, City of Issaquah